#### Massachusetts Firefighting Academy





Liquefied Petroleum Gas (LPG)

#### Objectives

- > Discuss propane and its properties
- > Show examples of common D.O.T. Cylinders
- > Discuss overfilling of 20lb cylinders
- Discuss overfill protection/safety changes

#### Objectives

- > Common A.S.M.E. Containers
- > Propane in transportation
- > Discuss BLEVE's and how to anticipate them
- Discuss CGI's and metering considerations


Pre - Test	
Identification	
Propane (LPG) 1075	

Properties	
Composition Of LPG	
LPG is composed of both propane and butane	
Source	
LPG is a by-product of oil refineries	

Toxicity  > Propane is non-toxic  > However, it is an asphyxiant  > It will displace the oxygen in the air  Specific Gravity  > .509  > Water is assigned the value of 1.0  > Any liquid with a specific gravity less than 1 will FLOAT on water  > Any liquid with a specific gravity greater than 1 will SINK in water  Therefore propane in its liquid form would	Properties  COLOR  Propane is colorless  ODOR  Propane is odorless  Propane is odorized by adding Mercaptan	
> .509 > Water is assigned the value of 1.0 > Any liquid with a specific gravity less than 1 will FLOAT on water > Any liquid with a specific gravity greater than 1 will SINK in water	<ul><li>&gt; Propane is non-toxic</li><li>&gt; However, it is an asphyxiant</li></ul>	
float on water	<ul> <li>&gt; .509</li> <li>&gt; Water is assigned the value of 1.0</li> <li>&gt; Any liquid with a specific gravity less than 1 will FLOAT on water</li> </ul>	

#### Weight

- > Liquid propane weighs approximately 4.4 pounds per gallon
- In comparison, water weighs approximately 8.3 pounds per gallon

#### Water Soluble

- > Is propane water soluble?
- > No, it will not mix with water

#### Vapor Density

- $\gt 1.6$  it is heavier than air since:
  - > Air is assigned the value of 1.0
  - > Any gas less than 1.0 is lighter than air and will rise
  - Any gas more than 1.0 is heavier than air and will collect in low spots such as ground level or basements, etc.


#### **Expansion Rate**

- > Thus, storing and transporting propane as a liquid is more economically sound
- > Example: one 10,000 gallon road transport full of liquid equals 270 transports full of vapor

Tem	nerat	tures
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#### Boiling Temperature > - 44° F

- Ignition Temperature
  > 920° F to 1120° F
  > Impurities in the propane may affect the ignition temperature

#### Flammable Range

- > 2.2% to 9.5%
- > Can be rounded off to 2% to 10%
- > What if the atmosphere is less than 2%?
- > Said to be too lean
- > What if the atmosphere is more than 10%?
- > Said to be too rich
- > Which would be potentially more dangerous?

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Flame Spread  > Approximately 900 feet per minute > Similar to Gasoline	
Storage Temperature  > Propane is stored at ambient temperature > Ambient temperature is the temperature of the day	
Storage Pressure  > 120 PSI at 70°F  > At higher temperatures the pressure will be greater  > At lower temperatures the pressure will be less	

#### Transport Pressure

- > 120 PSI at 70°F
- > Placing LPG in transport does not change the pressure
- > Temperature is the factor affecting pressure

#### LPG Vapor Pressures

<u>Temperature</u>	<u>Propane</u>	<u>Butane</u>
- 44° F	0	0
0° F	24	0
32° F	54	0
70° F	120	31
100° F	187	59
130° F	286	97

#### Specific Dangers

- > Flammable
- > Explosive in confined spaces
- > Asphyxiant will displace oxygen
- > Frostbite


Extinguishment  AGENT  Dry chemical agent  METHOD  Stop the flow of gas	
	•
Review Pretest	
Video Portable Extinguishers	

### Common D.O.T. Cylinders > One pound > Hand torches, small camping appliances > 20 pound > Recreational vehicles, grills, torches > 33 - 43 pound Industrial trucks (forklift / zamboni) may run on either liquid OR vapor Common D.O.T. Cylinders > 100 pound > Residential, usually in pairs, tar kettles > 200 pound to 400 pound > Residential or commercial applications Common D.O.T. Containers

1,000 - 14,000 Gallon

30,000 - 40,000 Gallon > Rail transports

> Bobtail delivery and Road transports

# Safety Relief Valve Operating Pressure: D.O.T. Cylinders - 375 PSI Note: At 160° F the Vapor Pressure of Propane is 390 PSI The 20 Lb Cylinder / Gas Grills Cylinder Overfilling

Cylinder Overfilling	
At 80% full and at 60° F, a 20 LB. cylinder must be heated to 160° F for anything to	
happen to it.	
Cylinder Overfilling	
At 95% full and at 60° F, a 20 LB. propane cylinder would need only a 22°	
temperature increase to become liquid full.	
	]
Cylinder Overfilling	
At 99% full and at 60° F, a 20 LB. cylinder	
would need only a 4.4° F temperature rise to cause a pressure great enough to	
operate the pressure relief valve.	

#### Safety Changes

- > Changes to gas grills
- > Changes to D.O.T. cylinders

#### Safety Changes

Gas Grills Quick Release Coupling



#### Adaptor With QRC

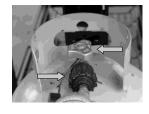


# Adaptor





New Valve Assembly



Overfill Protection Device (OPD)	
Video Overfill Protection Device	
Common A.S.M.E. Containers  500 - 5,000 Gallon  Normally found in commercial applications  10,000 Gallon and above  Storage facilities	

Storage Facilities		
Safety Relief Valve Operating Pressure: A.S.M.E. Containers - 250 PSI		
	•	
LPG in Road Transportation		

#### The Bobtail Delivery Truck





#### **Bobtail Emergency Shutoffs**



- > Manual shutoff behind cab on drivers side of truck
- > In the event of an emergency will close the main liquid discharge valve

#### **Bobtail Emergency Shutoffs**



Cable runs to rear of tank

#### **Bobtail Emergency Shutoffs**



- Newer units are equipped with remote control shut off
- > Device is activated by a garage door type control kept with the driver

#### **Bobtail Emergency Shutoffs**



Fusible link in cable can also shut down liquid valve in the event of a fire

#### LP Gas Transports



Emergency Shut Offs	
Front Rear	
B.L.E.V.E.  > Boiling > Liquid > Expanding > Vapor > Explosion	
Video Bleve Update	

## The Vapor Space Is the Danger Area > Cool the vapor space of a heated cylinder > Shut the gas off by the control valve if possible If the flow of burning gas cannot be shut off, allow the propane cylinder to burn itself out When to Anticipate a BLEVE > Activation of PRV > Sounds from PRV increase > Pitch from PRV becomes higher or louder > Space between flame and PRV increases > Water hitting the tank turns to steam Combustible Gas Indicators

Combustible Gas Indicators  CGI's, also referred to as "explosive meters" or "explosimeters," are used to test atmospheres that may contain a sufficient concentration of combustible vapors to cause an explosion or support combustion	
Combustible Gas Indicators	
There are three different scales used on various CGI models:  > Percentage of lower explosive limit (LEL)  > Percentage of gas in air  > Parts Per Million (PPM)  The most common is the percentage of LEL meter	
CGI Response	
A properly set low level alarm on a CGI meter is 10% of the LEL for the calibration gas	
> The reason this percentage is fairly low is that it serves as a safety factor	

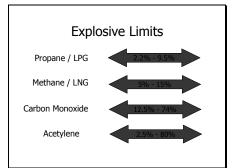
# CGI's and Oxygen > Oxygen concentrations will effect meter readings Oxygen Meters > Oxygen meters are used to detect the percentage of oxygen in atmosphere Most oxygen-sensing devices are calibrated to indicate concentrations between 0% and 25% Instrument Operation

### Combustible Gas Indicators > If a meter reading is 50% LEL, this would be equivalent to 2.5% vapor in air Too Lean FR 5-15% Too Rich 0 % 1 50% of lower explosive limit 100 % 2.5% volume in air Carbon Monoxide (CO) and Hydrogen Sulfide (H<sub>2</sub>S) Meters > These instruments utilize a detector that operates by chemical reaction with the gas > Like the oxygen meter, these meters are subject to interference from other gases or vapors CGI All CGI readings are relative to a calibration gas. When measuring another gas or vapor, the instrument still responds to the increased

temperature of the filament

#### **Conversion Factors**

Combustible Gas/ Vapor	Correction factor when Instrument is calibrated on Propane	Correction factor when Instrument is Calibrated on Methane
Hydrogen	0.61	1.11
Methane	0.55	1.0
Propane	1.0	1.82
N-Butane	1.0	1.82
N-Pentane	1.22	2.22
Methanol	0.65	1.18
Ethanol	0.85	1.54
Ammonia	0.46	0.83
Toluene	1.57	2.86
Gasoline	0.85	1.54



#### Summary

- Vapors are HEAVIER than air and will collect in low spaces
   Liquid leaks are 270 times worse than Vapor leaks
- $\,\,{\mbox{\tiny >}}\,\,\mbox{CGI's}$  are the only way to tell where the vapors truly are
- > LPG is found EVERYWHERE!