

**Sediment Toxicity
of
Petroleum Hydrocarbon Fractions**

*Prepared
for*

**Massachusetts Department of Environmental Protection
Office of Research and Standards
1 Winter Street
8th Floor
Boston, MA 02108**

*Prepared
by*

**BATTELLE
397 Washington Street
Duxbury, MA 02332**

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EXECUTIVE SUMMARY

There is a recognized need by regulatory agencies, such as the Massachusetts Department of Environmental Protection (MADEP) for the development of petroleum cleanup goals in sediment that are protective of aquatic receptors. This is due to the close proximity of many petroleum-contaminated sites to marine or fresh surface waters where petroleum products that have been released to the environment may reach the surface water and sediment, adversely affecting aquatic receptors. The first step toward developing cleanup goals is identifying toxicity-based sediment benchmarks.

The proposed approach for the development of sediment benchmarks is based on a methodology that divides the individual hydrocarbons from petroleum into several classes or groupings of compounds referred to as fractions, with similar chemical and toxicological properties. Four aliphatic and four aromatic fractions were chosen for the development of sediment benchmarks for the protection of aquatic receptors. Current MADEP analytical methods for petroleum in sediments, however, divide the hydrocarbons into different fractions based on human health risk. It is recommended that the existing MADEP methods be closely reviewed and method modifications be considered and evaluated, which will enable revised versions of those methods to be applied to sediment characterization for ecological risk.

The approach described in this report for estimating the acute and chronic toxicity of hydrocarbon fractions to aquatic organisms is based on equilibrium partitioning (EqP) theory. This theory states that the toxicity of hydrocarbons in sediments to benthic organisms is caused by the hydrocarbons that partition from the organic fraction of sediment particles into porewater and from porewater into the tissues of sediment-dwelling organisms. Both bioaccumulation and toxicity of hydrocarbons increase as the octanol-water partition coefficient (K_{ow}) of the hydrocarbon increases. A regression of toxicity data versus $\text{Log } K_{ow}$ produces a straight line from which toxicities of other hydrocarbons can be estimated if their $\text{Log } K_{ows}$ are known. Using the final chronic aquatic toxicity value, the sediment organic carbon/water partition coefficient (K_{oc}), and the fraction of organic carbon in sediment, equilibrium partitioning sediment benchmarks were derived for the four aliphatic and four aromatic hydrocarbon fractions of petroleum (Table ES-1). For comparison, sediment benchmarks were also derived based on the carbon fractions currently used by MADEP (Table ES-2).

Table ES-1. Sediment Benchmarks for Recommended Petroleum Hydrocarbon Fractions

Hydrocarbon Fraction	Geometric Mean Log K_{ow}	K_{oc}	Final Chronic Value ($\mu\text{g/L}$)	Sediment Benchmark (mg/kg oc)	Sediment Benchmark ($f_{oc} = 0.001$) (mg/kg)
<i>Aliphatic Hydrocarbons</i>					
$C_5 - C_8$	4.12	7.24×10^3	218	1591	1.59
$C_9 - C_{12}$	6.01	4.37×10^5	6.3	2722	2.72
$C_{13} - C_{18}$	8.57	1.10×10^8	0.05 ^a	5543	5.54
$C_{19} - C_{36}$	11.64	8.32×10^{10}	0.0001 ^a	9883	9.88
<i>Aromatic Hydrocarbons</i>					
$C_6 - C_8$	2.82	4.47×10^2	1191	531	0.53
$C_9 - C_{12}$	3.94	4.90×10^3	46.2	228	0.23
$C_{13} - C_{15}$	4.67	2.40×10^4	5.2	125	0.13
$C_{16} - C_{24}$	5.9	3.39×10^5	0.12 ^a	40	0.04

^a The fraction is not likely toxic because mean LC_{50} exceeds mean aqueous solubility.

Table ES-2. Sediment Benchmarks for Current MADEP Petroleum Hydrocarbon Fractions

Hydrocarbon Fraction	Geometric Mean Log K_{ow}	K_{oc}	Final Chronic Value ($\mu\text{g/L}$)	Sediment Benchmark (mg/kg oc)	Sediment Benchmark ($f_{oc} = 0.001$) (mg/kg)
<i>Aliphatic Hydrocarbons</i>					
$C_5 - C_8$	4.12	7.24×10^3	218	1591	1.59
$C_9 - C_{18}$	7.32	7.41×10^6	0.4	3167	3.17
$C_{19} - C_{36}$	11.64	8.32×10^{10}	0.0001 ^a	9883	9.88
<i>Aromatic Hydrocarbons</i>					
$C_9 - C_{10}$	3.84	3.98×10^3	59.4	236	0.24
$C_{11} - C_{22}$	4.81	3.31×10^4	2.8	92	0.09

^a The fraction is not likely toxic because mean LC_{50} exceeds mean aqueous solubility.

Various uncertainties in using the equilibrium partitioning approach to develop sediment benchmarks for petroleum are presented. These include the various methods for estimating K_{ow} and the resulting K_{ow} values; the limited amount of toxicity data for aliphatic hydrocarbons; the wide range of aromatic hydrocarbon toxicity data for both marine and freshwater species, as well as various test durations; and the selection of the most appropriate application factor, which is applied to account for differences in acute and chronic toxicity values and species sensitivity. These uncertainties were addressed in an appropriately conservative manner consistent with previous work by Hansen et al. (2003), DiToro et al. (1991), Total Petroleum Hydrocarbon Criteria Working Group (TPHCWG) (1997), and the European Community (EC) (2003).

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ACRONYMS AND ABBREVIATIONS

ACR	acute-to-chronic ratio
AF	application factor
BTEX	benzene, toluene, ethylbenzene, xylene
EC	European Community
ECETOC	European Center for Ecotoxicology and Toxicology of Chemicals
EPH	extractable petroleum hydrocarbons
EqP	equilibrium partitioning
ESB	equilibrium partitioning sediment benchmark
FACR	final acute-to-chronic ratio
FCV	final chronic value
f_{oc}	fraction of organic carbon in sediment
GC/FID	gas chromatography/flame ionization detection
GC/MS	gas chromatography/mass spectrometry
IR	infrared spectroscopy
K_{oc}	organic carbon-water partition coefficient
K_{ow}	octanol-water partition coefficient
LC ₅₀	lethal concentration required to cause mortality to 50% of test organisms
LOEC	lowest observed effect concentration
MADEP	Massachusetts Department of Environmental Protection
mg/L	milligrams per liter
mM/L	millimoles per liter
NAPL	non-aqueous phase liquid
NOEC	no observed effect concentration
PAH	polycyclic aromatic hydrocarbon
PID	photoionization detection
ppb	parts per billion
TPH	total petroleum hydrocarbons
TPHCWG	Total Petroleum Hydrocarbon Criteria Working Group
TPH-d	TPH-diesel range
TPH-g	TPH-gasoline range
TPH-r	TPH-residual range
µg/L	micrograms per liter
US	United States
USEPA	United States Environmental Protection Agency
UST	underground storage tank
VPH	volatile petroleum hydrocarbons

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1.0 INTRODUCTION

Products derived from refining of crude oil are among the most abundant consumer products in the world. The average consumption of crude oil in the United States (U.S.) is about 714 million gallons/day (Minerals Management Service, 1991; Beck, 1996). About two-thirds of the total crude oil is refined for use as fuel in land vehicles, airplanes, and ships. The consumption of automotive gasoline is approximately 327 million gallons/day (Beck, 1996). About 190 million gallons/day of middle distillate fuels (jet fuel, kerosene, diesel fuels, and home heating oils) and about 48 million gallons/day of residual oil (bunker fuels) are consumed in the U.S. The remainder of the crude oil consumed each day is used for other non-fuel distillate (*e.g.*, lubricating oils) and residual products (*e.g.*, road asphalt), as well as feed stocks for the petrochemical industry.

A small fraction of the petroleum products consumed in the U.S. is released to land, either accidentally or intentionally. Release of refined oils, particularly gasoline, from leaking underground storage tanks (USTs) is the most widely recognized source of petroleum contamination of soils and groundwater (Dowd, 1984; Murphy et al., 1987). Refined oil products also may be released to the land from leaky pipes, during transfer to and from storage tanks, by application of waste oil to dirt roads to control dust, and by land disposal of waste crankcase and other oil products.

It was recognized in the 1970s and 1980s that petroleum hydrocarbons can migrate from petroleum-contaminated soils and adversely affect terrestrial and aquatic ecosystems and humans. Existing cleanup goals for petroleum in soil and groundwater focus on the protection of human health. Because of the close proximity of many petroleum-contaminated sites to marine or fresh waters, the potential exists for petroleum products that have been released to the environment to reach the surface water and adversely affect aquatic receptors. Appropriate cleanup goals need to be established for petroleum hydrocarbons in sediments before remedial actions can be initiated. The level (hydrocarbon concentration) to which a site's sediment must be cleaned should be set low enough to protect the environment, but not necessarily to represent pristine conditions. Certainly, these cleanup goals must be acceptable and consistent with State and Federal environmental regulations.

1.1 Objectives

Oil contamination is being recognized with increasing frequency as a major contributor to the hazard to aquatic life in contaminated sediments, particularly in areas of intense anthropogenic activity (Neff, 1979; 2002). However, because refined petroleum products are complex variable mixtures of hundreds or thousands of organic chemicals, it has been extremely difficult to develop cleanup benchmarks for total petroleum hydrocarbon (TPH)¹ concentrations in sediments that are acceptable to both regulatory agencies and regulated entities. Consequently, there is a recognized need by regulatory agencies, such as the Massachusetts Department of Environmental Protection (MADEP) for the development of petroleum cleanup goals that are protective of aquatic receptors. Thus, the intent of this report is to develop scientifically defensible, risk-based cleanup levels for petroleum for the protection of aquatic receptors by incorporating and modifying applicable elements of current scientific approaches from Hansen et al. (2003), DiToro et al. (1991), and the Total Petroleum Hydrocarbon Criteria Working Group (TPHCWG) (1997).

¹ TPH is a generic term used to define the sum of hydrocarbon concentrations occurring in crude and refined petroleum that yield a positive hydrocarbon signal when analyzed by a specific analytical method.

1.2 Petroleum Hydrocarbons

1.2.1 Composition

Crude petroleum is an extremely complex mixture of fossil material, primarily of plant origin (Speers and Whitehead, 1969). It contains thousands of organic, and a smaller number of inorganic, compounds. Crude oils vary tremendously in chemical composition, in relative concentrations of different chemicals, and in physical properties; no two are alike. Crude oils are refined to produce a wide variety of refined and residual products, mostly fuels that contain a smaller number of chemicals that are usually within a defined boiling point range.

A particular crude oil may contain organic compounds ranging in molecular weight from methane (molecular weight 16), a gas at room temperature and pressure, to complex polymeric structures, such as asphaltenes with molecular weights up to at least 100,000 (Kallio, 1976). Hydrocarbons, organic chemicals composed solely of carbon and hydrogen, are by far the most abundant chemicals in crude and refined oils. Variable amounts of organic chemicals containing sulfur, nitrogen, or oxygen also are present in all crude and some refined oils.

The hydrocarbons in petroleum are aliphatic (saturated), aromatic (unsaturated), or a combination of both (Figure 1). Aliphatic hydrocarbons, also called alkanes or paraffins, are composed of chains of carbon atoms linked by single covalent bonds. Chemical bonds not occupied by carbon-carbon bonds are occupied by hydrogen atoms. Aliphatic hydrocarbons in petroleum may be normal (a linear chain), branched, or cyclic. They may range in size from methane (CH_4) to at least C_{78} (a chain of 78 carbons). Some refined oils, particularly light fuels, such as gasoline and kerosene, may contain olefins (aliphatic hydrocarbons containing one or more carbon-carbon double, or occasionally triple bonds) generated during the refining process. Olefins usually represent less than a few percent of the hydrocarbons in fuels.

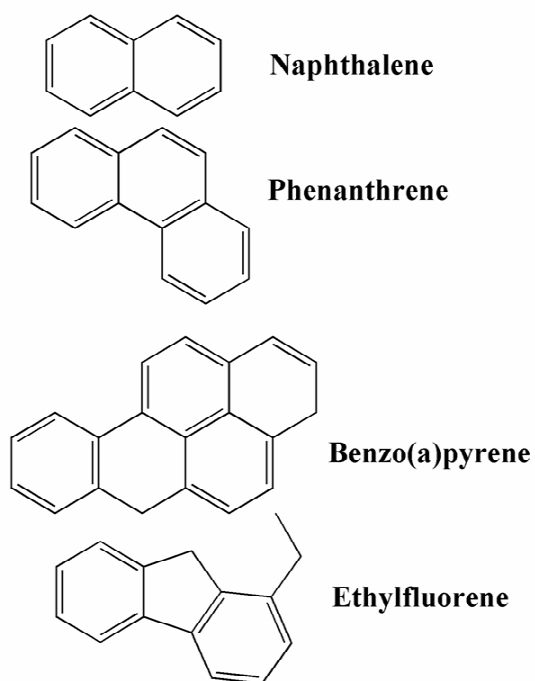
Aromatic hydrocarbons in petroleum are composed of one or more benzene rings, a six-carbon ring containing nine equally shared carbon-carbon covalent bonds. Each carbon atom in benzene is bonded to one hydrogen atom, which can be substituted by a methyl ($-\text{CH}_3$), ethyl ($-\text{CH}_2\text{CH}_3$), or longer-chain aliphatic group(s) called an alkyl group(s). Benzene and alkyl benzenes with one or two methyl or ethyl groups (toluene, ethylbenzene, and *m*-, *p*-, and *o*-xylene) are the most abundant aromatic hydrocarbons in most crude and refined oils.

Benzene may be linked to other benzene molecules by single covalent bonds to form compounds such as biphenyl and terphenyl. More frequently, two or more benzenes are fused (sharing two carbons) to form polycyclic aromatic hydrocarbons (also called polynuclear aromatic hydrocarbons [PAHs]) (Neff, 1979). Naphthalene, composed of two fused benzene rings, has a molecular weight of 128.2 and is the smallest PAH. Coronene, with six condensed, fused rings and a molecular weight of 300.3 is the highest molecular weight PAH considered to have sufficient environmental mobility (aqueous solubility of approximately 0.1 $\mu\text{g/L}$: parts per billion [ppb]) to be potentially toxic.

A large number of products, mostly fuels, are produced from crude oil by refining. Refining involves distillation to isolate oil fractions containing hydrocarbons with different boiling points and stimulation of chemical reactions that convert hydrocarbons in the fractions from one form to another. Refined oil products include gasoline and middle distillate fuels, such as diesel fuel, jet fuel, kerosene, and home heating oil. The petroleum fraction remaining after removal of light and middle distillate fractions is called residual oil, which is used to fuel ships and power plants (bunker fuel) or to produce road paving asphalt. Lubricating oils and petroleum tars also are

made from residual oil. Refined and residual petroleum products contain all the chemical classes present in crude petroleum, as well as some compounds produced during refining or added to the finished product to improve its properties (Nyer and Skladany, 1989; King, 1992). Each refined or residual product contains primarily crude oil chemicals over a specified boiling point range. Gasoline has an approximate boiling point range from 40 to 205°C; middle distillate fuels have an approximate boiling point range between 175 and 375°C; residual fuel oils are blended from crude oil fractions generally boiling between about 350 and 700°C.

Polycyclic Aromatic Hydrocarbons



Aliphatic Hydrocarbons

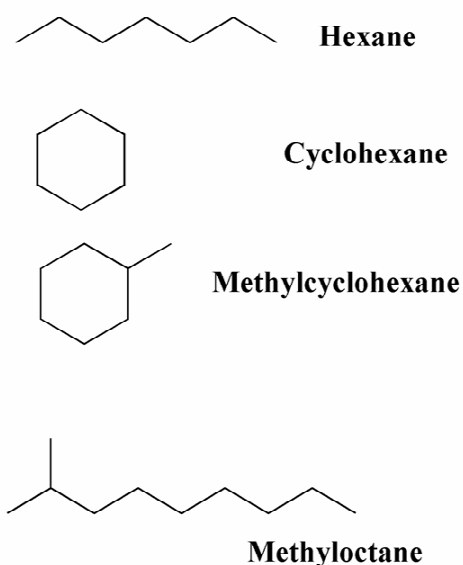


Figure 1. Molecular Structure of Several Aromatic and Aliphatic Hydrocarbons Found in Petroleum.

1.2.2 Fractionation

Crude and refined petroleum contain hundreds (gasoline) to thousands (middle and heavy fuels and crude oil) of hydrocarbons and related organic compounds that vary widely in persistence in the environment and toxicity. The relative concentrations of different hydrocarbons vary widely in different crude, refined, and residual oil products and hydrocarbon composition changes rapidly following release to the environment (see Section 2.1.1). Therefore, estimation of ecological risk of petroleum products in sediments and groundwater, based on a single TPH value, is an over-simplification, and often contributes to large errors in the risk estimation. Determining risk based on the concentrations of the tens and possibly thousands of individual hydrocarbons in the petroleum contamination is overly complex and costly. The proposed approach for development of sediment cleanup goals is based on a methodology that divides the hydrocarbons into several classes or groupings of compounds referred to as fractions, with similar chemical and toxicological properties. This greatly improves the ability to more accurately

estimate the potential for ecological risk of the weathered petroleum product in sediments and groundwater, compared to a method based only on TPH data.

As a general rule, the aqueous solubility of hydrocarbons decreases and aquatic toxicity increases with increasing molecular weight (number of carbons). Aliphatic hydrocarbons have a lower solubility and toxicity than aromatic hydrocarbons with the same number of carbons. Thus, for risk assessment, the aliphatic and aromatic hydrocarbons in oil-contaminated environmental samples can be divided into several fractions, each covering a limited range of carbon numbers with similar solubility and toxicity. Each fraction, except for the two largest carbon-number fractions, contains hydrocarbons with a small carbon-length range so that the hydrocarbons in each fraction will have a small range of physical/chemical properties and toxicity. Four aliphatic and four aromatic fractions were recommended and used in this method for the development of sediment benchmarks for the protection of aquatic receptors. Table 1 presents the recommended fractions for this method as well as the current MADEP fractions used to characterize human health risk.

Table 1. Comparison of Different Hydrocarbon Fractions to Characterize Risk

Aliphatic Hydrocarbons		Aromatic Hydrocarbons	
<i>Recommended Fractions</i>	<i>Current MADEP Fractions</i>	<i>Recommended Fractions</i>	<i>Current MADEP Fractions</i>
C ₅ -C ₈	C ₅ -C ₈	C ₆ -C ₈ (Benzene, Toluene, Ethylbenzene, Xylene) ^a	
C ₉ -C ₁₂	C ₉ -C ₁₈	C ₉ -C ₁₂	C ₉ -C ₁₀
C ₁₃ -C ₁₈		C ₁₃ -C ₁₅	C ₁₁ -C ₂₂
C ₁₉ -C ₃₆	C ₁₉ -C ₃₆	C ₁₆ -C ₂₄	

^a The BTEX compounds (which are C₆ through C₈ aromatic compounds) are not recommended for standard ecological risk assessment of sediments, but may be considered for contaminant characterization on a case-by-case basis. When characterized, it is recommended that the BTEX compounds, because of their high volatility, be determined by specific methods, and not C₆ through C₈ aromatics as a general fraction.

The recommended fractions in Table 1 group hydrocarbons having similar chemical characteristics and abundance in petroleum, similar toxicities to aquatic organisms, and similar physicochemical properties related to fate and transport. The C₉-C₁₂ aromatic fraction contains naphthalene (C₁₀) and mono-methyl- and dimethyl-naphthalenes (C₁₁ and C₁₂); the C₁₃-C₁₅ aromatic fraction contains fluorene (C₁₃), methylfluorene (C₁₄), phenanthrene/anthracene (C₁₄), methyl phenanthrene/anthracene (C₁₅), and tri- and tetra-methyl naphthalenes. The high molecular weight C₁₆-C₂₄ aromatic fraction contains four- and five-ring PAHs, including fluoranthene/pyrene (C₁₆) and higher molecular weight parent (unalkylated) and alkylated PAHs, including the carcinogen, benzo(a)pyrene (C₂₀). The highest molecular weight PAHs in the C₁₆-C₂₄ aromatic fraction have such low solubility that they have low mobility in sediments and porewater.

The fractions recommended for this method are similar to the fractions recommended by the TPHCWG (1997) and MADEP (2002) to estimate human health risks of TPH in sediments, with minor changes. The aromatic C₁₁-C₂₂ fraction recommended by MADEP is large and includes hydrocarbons from methyl-naphthalene to dibenz(a,h)anthracene. For this method, this fraction has been divided into two fractions, C₁₃-C₁₅ and C₁₆-C₂₄ (Table 1), based on the observation that saturated aqueous solutions of aromatic hydrocarbons larger than fluoranthene (C₁₆) have such low aqueous solubilities that they may not be acutely toxic to aquatic organisms (Di Toro et al., 2007).

Furthermore, the relatively volatile, and generally more water soluble aromatic compounds (such as benzene, toluene, ethylbenzene, and xylenes [BTEX]) are quite mobile in aquatic environments and are not expected to accumulate in sediments to concentrations that would pose a significant ecological risk. Aliphatic hydrocarbons in the C₅-C₈ aliphatic fraction have higher K_{ow} values (are less soluble) than aromatic hydrocarbons in the C₆-C₉ fraction, but are more biodegradable, reducing their persistence in sediments. However, the light aliphatic hydrocarbons in the C₅-C₈ and C₉-C₁₂ fractions are sufficiently soluble that they probably contribute to the toxicity of sediments contaminated with light and middle distillate fuels, which almost always contain much higher concentrations of aliphatic than aromatic hydrocarbons, unless the light aliphatic hydrocarbons have been depleted by weathering (Neff et al., 2000). The high molecular weight aliphatic hydrocarbons (aliphatic fractions C₁₃-C₁₈ and C₁₉-C₃₆) have such low solubilities and high log K_{ow}s (7.2 – 13.07) that they have a high affinity for the sediment organic carbon phase or oil phase in the sediments and little partitions into the water phase where it can be bioconcentrated by aquatic organisms. Thus, the high molecular weight aliphatic hydrocarbons probably do not contribute significantly to the chemical toxicity of oil-contaminated sediments.

1.2.3 Analytical Methods

Analytical methods available to quantify TPH in soils, sediments, and water include solvent extraction/gravimetry, infrared spectroscopy (IR), immunoassay, and gas chromatography/flame ionization detection (GC/FID). TPH is comprised of aliphatic and aromatic hydrocarbons, as well as other non-polar organic compounds in petroleum; a number of non-petroleum organic compounds may also contribute to the measured value, depending on the specificity of the method. The methods give different results for TPH in soils, sediments, and water, because they measure different groups of petroleum hydrocarbons and are subject to interferences from non-petroleum organic compounds, such as biogenic hydrocarbons, lipids, and waxes from decaying plant and microbial matter and other types of organic chemicals frequently found in soils, sediments, and groundwater. Depending on the physical and chemical properties of the environmental samples and the objectives of the analysis (quantitative for screening or qualitative and quantitative to support a risk assessment), the analytical measurement approach has a profound effect on the biases (over- or under-estimation of the true petroleum concentration) associated with the generated results. Because of the limitations of all the TPH analytical methods, there is a danger of making a risk-based decision with inaccurate or less-than-adequate information.

To remove some of the inherent problems, a first-level evaluation/interpretation of sample composition is critical. Interpretive errors commonly occur when a naturally occurring biogenic hydrocarbon signature (either as the sole hydrocarbon or in a mixed hydrocarbon assemblage) or a non-hydrocarbon present in the sample is mistakenly identified as part of a petroleum hydrocarbon mixture. All the methods tend to overestimate the true TPH concentration in environmental samples. There are many classic examples of “mistaken identity” in reports of TPH in soils, sediments, and water.

Gravimetry, IR, and immunoassay methods for TPH are not qualitative (they do not identify individual hydrocarbons) and, therefore, have limited applicability in risk assessment. Therefore, this discussion will focus on quantifying and characterizing TPH in environmental matrices by GC/FID. The four analytical methods for TPH analysis differ in ease and speed of performance, cost, and types and quality of results obtained. Another method, based on temperature-programmed capillary column gas chromatography with analyte identification and quantification by mass spectrometry (GC/MS), is the best method for differentiating between petroleum-derived and other (mostly biogenic or combustion sourced) hydrocarbons in environmental samples. However, because of its technical difficulty and high cost, it is rarely used in assessments at sites where TPH is the primary driver. GC/MS is particularly useful for analyzing specific compounds or classes of compounds, such as aromatic hydrocarbons. A high quality GC/FID is generally the preferred analytical approach for determining saturated hydrocarbons and TPH, as discussed below.

1.2.3.1 USEPA Method 8015

Gas chromatography methods, particularly USEPA Method 8015 modified for hydrocarbon analysis, are generally preferred for measuring bulk petroleum in environmental samples, and specifically saturated hydrocarbons. This is because of their specificity in product identification if most interferences are removed by suitable cleanup methods, and because of their greater accuracy than gravimetric and IR methods in quantification of petroleum hydrocarbons. Method 8015 uses GC/FID for the determination of TPH. This method is the most commonly used tool for the analysis of TPH in environmental samples. The technique is capable of measuring a wide range of hydrocarbon classes, from approximately C_6 through C_{44} (when optimized), although the reliability is lower for compounds with a molecular weight of less than C_9 ². The strength of the method is that it is both quantitative and qualitative and can therefore be used in risk assessments.

Method 8015 is used routinely as a first-level identification of both product type and weathering state. The GC/FID method is used to quantify different types of refined and residual petroleum in environmental samples, including TPH-gasoline range (TPH-g), TPH-diesel range (TPH-d), and TPH-residual range (TPH-r). These types of hydrocarbon assemblages are based on quantification of the area under chromatographic peaks in a certain segment of the gas chromatogram corresponding to hydrocarbons of a defined carbon number and boiling point range. The area under the chromatogram (including both aromatic and aliphatic hydrocarbons) between C_6 and C_{10} is defined as TPH-g; the area under the chromatogram between C_{10} and C_{24} is defined as TPH-d; and the area under the chromatogram $>C_{24}$ is defined as TPH-r.

Method 8015 is susceptible to misinterpretations of the chromatographic results, especially by inexperienced chemists. These misinterpretations can have dramatic effects on results of a site assessment or risk assessment. For example, unnecessary site cleanups of sediments have been performed due to the misinterpretation of chromatograms because the GC/FID signature of sediment often closely resembles that of a weathered fuel oil #6. Other common misinterpretations involve misidentification of biogenic materials, including some coals and some plant organic material containing high molecular weight components, such as plant waxes.

Unless rigorous cleanup of the sample extract is performed before analysis by GC/FID (not specifically required by USEPA), Method 8015 may give erroneously high values for TPH, due to interference from oil microdroplets (frequently neither in solution nor water soluble) or co-extracted non-hydrocarbon chemicals in the samples (Zemo, 1997). Interfering non-hydrocarbon

² The C_6 - C_8 (BTEX) compounds are best analyzed by a purge and trap technique, with the instrument fitted with an MS rather than an FID for greatest reliability.

compounds may co-elute from the GC column with hydrocarbons and be quantified as part of one of the petroleum product classes. Method 8015 provides tremendous flexibility for analytical laboratories with regard to the actual GC conditions for the analysis. To expedite analysis time, high GC oven rates often are used, resulting in poor resolution of compounds and a greater propensity to misidentify non-hydrocarbons as hydrocarbons.

Method 8015 is used in most remedial investigations and risk assessments at petroleum-contaminated sites in the U.S. The method can provide useful data if its limitations are clearly understood; with proper extract preparation, GC analytical conditions, and interpretation, Method 8015 can be adequate for TPH.

1.2.3.2 MADEP EPH and VPH Method Adoption and Recommendations

The proposed analytical method for petroleum hydrocarbons will, as much as possible, be based on the analytical approach described in the MADEP method documents *Method for the Determination of Extractable Petroleum Hydrocarbons (EPH)* and the *Method for the Determination of Volatile Petroleum Hydrocarbons (VPH)* (MADEP, 2004a;b). The MADEP EPH method may be considered a modified version of EPA Method 8015, because it is a GC/FID based analysis of extractable petroleum hydrocarbons. The MADEP VPH method is also GC/FID based (in conjunction with photoionization detection [PID]). However, the MADEP methods are notably enhanced from the USEPA method by employing a fractionation scheme for separating the petroleum hydrocarbons into multiple aliphatic and aromatic molecular size and boiling point fractions prior to final determination on the analytical instrument. This generates additional information for improved risk analysis and other site cleanup decisions. The MADEP methods are also enhanced by including better sample purification methods than those described in Method 8015, which results in more specific and reliable data.

Currently, in the MADEP methods, the hydrocarbon fractions span a wide range of carbon numbers and physical/chemical and toxicological properties. The fractions recommended in this guidance are somewhat similar to the fractions already used by MADEP in soil assessment (*Characterizing Risks Posed by Petroleum Contaminated Sites: Implementation of MADEP VPH/EPH Approach* MADEP, October 31, 2002), which are generated using the MADEP EPH and VPH analytical methods. However, the carbon fractions described in this document have been modified to be more appropriate for the management of petroleum risk in aquatic sediment, and, consequently, the MADEP EPH and VPH analytical methods cannot be directly applied.

A review of the recommended fractionation scheme in Table 1, and the fractions described in the referenced MADEP analytical method and risk characterization documents, indicates some overlap, particularly with the aliphatic hydrocarbons, and some components with less in common. It is recommended that the existing MADEP methods be closely reviewed and method modifications be considered and evaluated, which will enable revised versions of those methods to be applied to sediment risk characterization. Method modifications that are recommended should be considered for evaluation, and a revised method should be incorporated if methods are successful. Recommendations include the following:

- Concurrent analysis of the three aromatic fractions (C₉-C₁₂, C₁₃-C₁₅, and C₁₆-C₂₄) as part of a revised EPH method.
- Concurrent analysis of the three highest molecular weight aliphatic fractions (C₉-C₁₂, C₁₃-C₁₈, and C₁₉-C₂₄) as part of a revised EPH method.
- The aliphatic C₅-C₈ would still require the VPH analysis. However, we recommend that the inclusion of this fraction be considered on a site-by-site basis, and, if appropriate, be

excluded from the chemical analysis and risk characterization. The elimination of C₅-C₈ aliphatic fraction would mean sampling and analytical cost savings, since this fraction would require separate handling and analysis.

- It is recognized that there would likely be greater loss of the C₉-compounds than most other compounds, using a revised EPH method that would include the C₉s. However, it is expected that these compounds will contribute less to the risk, on a relative basis, than the other measured petroleum related compounds, and the benefit of incorporating the C₉s into the first fraction of a revised EPH method would likely outweigh the drawbacks.

The suggested method revisions will require method development and validation work by a laboratory experienced in developing and modifying analytical methods. Further adjustments to these initial technical assumptions may be required, based on the outcome of the initial method development and validation work. A revised analytical method would then be written. MADEP may then wish to conduct a round robin study, as part of a method validation and refinement process, and then finalize the method.

Although analytical methodology changes are needed to fully support the ecological risk assessment objectives of this project, the method modifications will likely not be substantial from a technical perspective. It is anticipated that the new method can be based on the existing MADEP method(s), with some potential modifications to components such as (1) solvent elution of chromatography columns, (2) sample extract concentration temperatures and techniques, (3) GC temperature program, and (4) selection of compounds for fraction markers. It is also anticipated that any laboratories that are currently employing the MADEP VPH and EPH methods will be able to master the new method developed for the ecological risk assessment of sediments.

2.0 CONCEPTUAL SITE MODEL

2.1 Fates of Petroleum Products

Following release to the environment, refined and residual petroleum products may accumulate in soils and sediments where they undergo several dispersal and weathering processes that affect the composition and toxicity of the hydrocarbon mixtures (Neff et al., 1994). The most important weathering processes for refined and residual petroleum include dispersion, evaporation, dissolution, and biodegradation.

When released in large volumes to aquatic environments and sediment, refined and residual petroleum products tend to retain their identity as a separate oil phase (*i.e.*, nonaqueous-phase liquid [NAPL]). The rate and extent of migration of a NAPL into and through soil, sediment, and groundwater depends on the viscosity, density, and interfacial tension of the oil, and the permeability and porosity of the medium (Sale et al., 1992).

Volatile hydrocarbons, such as those in the lowest carbon fractions of aliphatic and aromatic hydrocarbons (Table 1) evaporate rapidly from the surface of sediments and the overlying water column. They evaporate slower from deep sediment layers, particularly if the sediments are fine-grained (silts and clays) with low permeability. Hydrocarbons are lost from buried sediment layers mainly by dissolution or dispersion in water percolating through the oiled sediment layer, or by microbial degradation. If the subsurface oil NAPL comes in contact with surface water or infiltrating water, hydrocarbons will dissolve from the oil phase into the water phase in direct relation to their relative solubilities in the two phases. Generally speaking, the aqueous solubility of petroleum hydrocarbons is inversely proportional to molecular weight, and aliphatic hydrocarbons are less soluble than aromatic hydrocarbons of similar molecular weight. Thus, low molecular weight aromatic hydrocarbons and, to a lesser extent, lower molecular weight aliphatic hydrocarbons tend to dissolve slowly from the petroleum-contaminated sediment or NAPL into the porewater and are carried away from the oil deposit with porewater or surface water flow (Figure 2).

Sediment bacteria and fungi may degrade petroleum hydrocarbons (Figure 2). Low molecular weight normal alkanes are degraded most rapidly, followed by branched alkanes and higher molecular weight normal alkanes. Low molecular weight, soluble aromatic hydrocarbons also may be degraded rapidly. PAHs and cyclic alkanes are more resistant to microbial degradation and degrade more slowly. Thus, as oil weathers, its composition changes with loss of low molecular weight aliphatic and aromatic hydrocarbons, and slower declines in concentrations of higher molecular weight hydrocarbons.

The effects of the combined weathering processes generally reduce the concentration of low molecular weight slightly soluble aliphatic and aromatic hydrocarbons in petroleum-contaminated sediment. Thus, the composition of a TPH fraction that has weathered for some time is quite different from that of the TPH fraction of the refined or residual petroleum product that was originally released. The relative toxicity of TPH fractions also changes, usually decreasing, during natural weathering (Di Toro et al., 2007). Because the compositions of TPH fractions of different refined and residual petroleum products vary widely and change during natural weathering, it is difficult to predict the concentration of TPH in sediment that does not pose a potential risk to the aquatic environment.

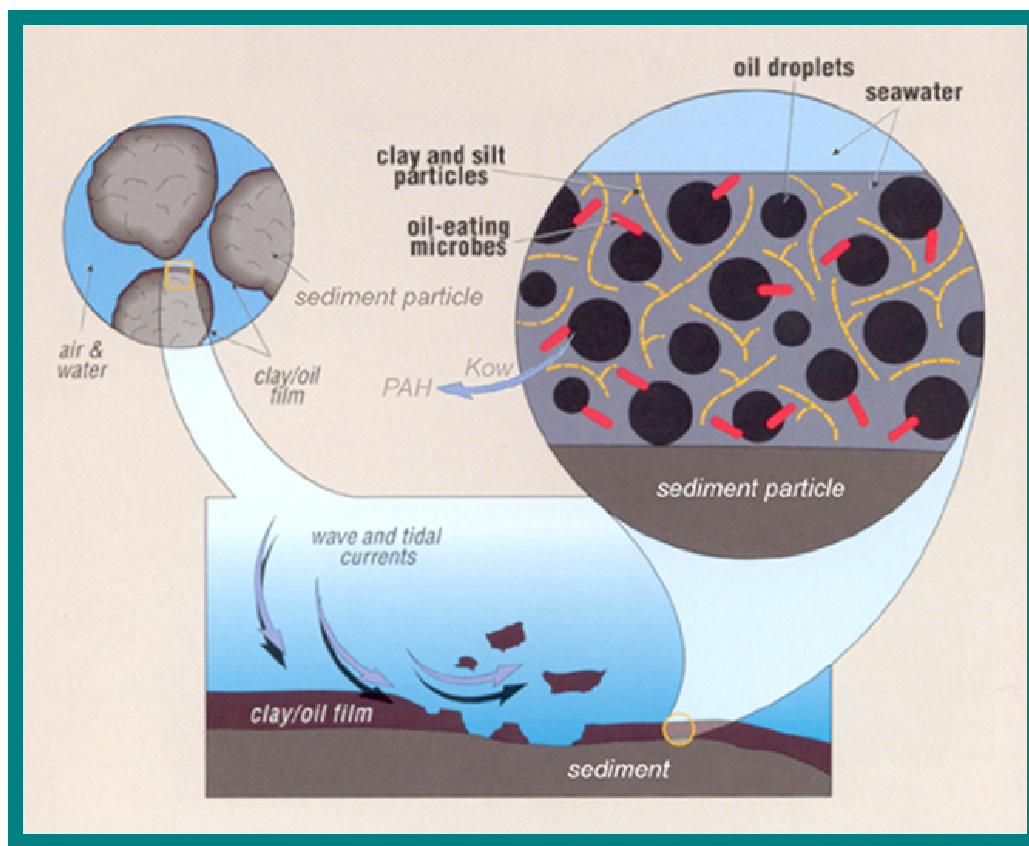


Figure 2. Partitioning of Hydrocarbons in Sediment

2.2 Equilibrium Partitioning

Toxic organic contaminants in bottom sediments of lakes, rivers, wetlands, and coastal waters create the potential for continued environmental degradation, even when water column contaminant levels comply with established water quality criteria (DiToro et al., 1991). Hydrocarbons have low aqueous solubilities and high affinities for adsorption to sediment organic matter. They also adsorb to dissolved and suspended organic matter and readily partition across permeable external (*e.g.*, gills) and internal (gut epithelium) membranes of water column and sediment dwelling organisms, where they accumulate in tissue lipids (Neff, 2002). Hydrocarbons in solution in sediment porewater are more bioavailable and toxic to sediment-dwelling organisms than hydrocarbons adsorbed to sediment particles, particularly combustion soot (Thorsen et al., 2004) or associated with a NAPL (*e.g.*, petroleum, creosote, or tar) (DiToro et al., 1991; Hansen et al., 2003; Neff et al., 2005).

Hydrocarbons associated with sediment particles are considered slightly bioavailable; bioaccumulation of particle-associated hydrocarbons by direct contact with external body surfaces or following ingestion of the particles is thought to involve an intermediate step in which the hydrocarbons desorb from the particle into the water and then are absorbed across an epithelium (Di Toro et al., 1991; 2000; Neff, 2002; Hansen et al., 2003). Hydrocarbons associated with sediment particles must partition from the sediment particles into solution before they can

move across biological membranes and be taken up by the organism. For low molecular weight hydrocarbons ($\log K_{ow} < 5.5$) with high solubility, sufficient levels of hydrocarbons are present in the porewater to allow rapid bioaccumulation across external permeable membranes or across the gut epithelium when particles and porewater are ingested. In the gut, “solubilization” of particulate hydrocarbons is aided by surfactants and enzymes secreted by the animal (Mayer et al., 1996). For high molecular weight hydrocarbons ($\log K_{ow} > 5.5$) with low solubility and high affinity for particles, few hydrocarbons are able to partition from the particles into bulk porewater. However, if hydrocarbon-contaminated sediment particles come into direct contact with permeable epithelia (e.g., gill, gut epithelium), some high molecular weight hydrocarbons may dissolve in the thin film of water between the particle and membrane surfaces and partition into the membrane. In the gut of fish, the natural surfactants are particularly important in facilitating this transfer. However, the assimilation efficiency for high molecular weight hydrocarbons in the gut of benthic invertebrates is low (generally $< 10\%$). The uptake of non-polar organic chemicals with $\log K_{ow} > 5$ by fish is almost exclusively from food or ingested sediments, while the uptake of chemicals with $\log K_{ow} < 5$ is primarily from the water (via the gills). The same is likely true for benthic invertebrates (Neff, 2002).

The best approach for estimating the toxicity of hydrocarbons in sediments to benthic organisms is to estimate the hydrocarbon concentration in sediment porewater and then compare the estimated concentration to water quality criteria for the hydrocarbon (Figure 3) as described in USEPA Guidance (DiToro et al., 1991; Hansen et al., 2003). The dissolved phase of hydrocarbons in sediment porewater can be estimated based on equilibrium partitioning theory (EqP) as described by Hansen et al. (2003). Equilibrium partitioning refers to the assumption that a state of equilibrium exists between the chemical sorbed to the particulate sediment organic carbon and the sediment porewater and between sediment porewater and the tissues of an organism in contact with the water (Figure 3). It makes several additional assumptions:

- 1) The partitioning of hydrocarbons between sediment organic carbon and porewater (interstitial water) and between porewater and tissues of aquatic organisms in contact with the porewater is at or near equilibrium;
- 2) The concentration in each phase (sediment, porewater, and tissue) can be predicted using appropriate partition coefficients and the measured or estimated concentration in the other phases;
- 3) The effective sediment exposure concentration of hydrocarbons to organisms is the same regardless of the exposure route (e.g., from sediment porewater via respiration or from sediment via ingestion)³;

³ Only the bioavailable forms of chemicals are toxic to aquatic organisms (Di Toro et al., 1991; Neff, 2002). A chemical is said to be bioavailable if it is in a form that can move through or bind to the surface membranes of an organism (e.g., skin, gill epithelium, gut lining, cell membrane). Thus, the exposure concentration of hydrocarbons to aquatic and sediment-dwelling organisms is the fraction of the total hydrocarbon in the ambient medium (including the gut) that is in a bioavailable form (dissolved in the water and in contact with a permeable membrane). It is assumed that particulate hydrocarbons (adsorbed to sediment particles or associated with an oil phase) must first desorb or partition from the solid or NAPL into the water before they can be bioaccumulated. Thus, the effective exposure concentration is the measured or predicted concentration of the hydrocarbon in solution in the water bathing external and internal permeable membranes of the aquatic organism. Partitioning of hydrocarbons from the solid phase into water may be facilitated in the gut of sediment-ingesting animals by the natural detergents secreted by the animal into the digestive tract.

- 4) Effects concentrations of hydrocarbons in sediments can be estimated using the organic carbon-to-water partition coefficient (K_{oc}) and effects concentrations in water; and,
- 5) The derived equilibrium partitioning sediment benchmark (ESB) is protective of benthic and water column organisms.

Thus, the toxicity of hydrocarbons in sediments to benthic organisms is caused by the hydrocarbons that partition from the organic fraction of sediment particles into porewater and from porewater into the tissues of the organisms (Figure 3). The sediment benchmark, which is protective of benthic organisms, can be estimated from the K_{oc} and the chronic aquatic toxicity value for each hydrocarbon fraction.

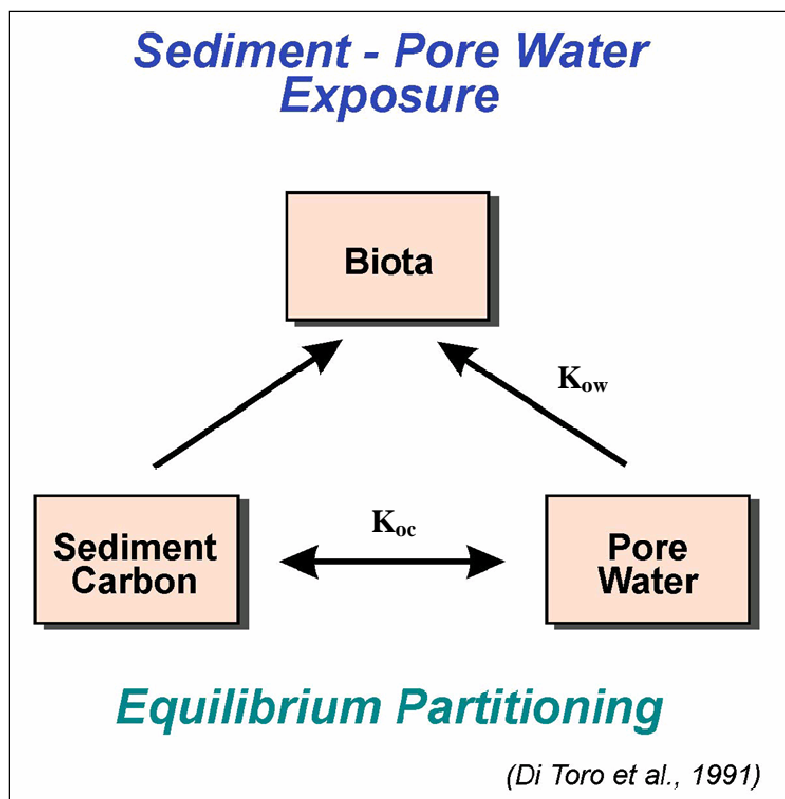


Figure 3. Equilibrium Partitioning Conceptual Site Model

2.3 Toxicity

For non-polar organic chemicals, such as petroleum hydrocarbons, properties such as water solubility, sorption to sediment particles, bioconcentration, and toxicity are proportional to hydrophobicity, which is expressed as the octanol-water partition coefficient (K_{ow}).⁴ Because octanol is a good surrogate for tissue lipids of aquatic organisms, K_{ow} can be used to predict the

⁴ K_{ow} is an expression of the relative affinity of a chemical for dissolution in the octanol and water phases of a 50/50 octanol/water mixture. It is measured as the ratio of the concentration of the chemical in the octanol phase to its concentration in the aqueous phase after equilibration.

bioconcentration of hydrocarbons from water into the lipid-rich tissues of aquatic organisms (Connell, 1993; McCarty et al., 1992; Di Toro et al., 2000). Toxicity depends on attainment of a critical concentration of the chemical in the tissues of an aquatic organism (McCarty et al., 1992); therefore, K_{ow} can also be used to predict the concentration of hydrocarbons in the porewater that is toxic to aquatic organisms. Both bioaccumulation and toxicity of hydrocarbons increase as Log K_{ow} increases. A regression of toxicity data versus Log K_{ow} produces a straight line from which acute toxicities of other hydrocarbons can be estimated if their Log K_{ow} s are known. This is the approach described in this report for estimating the acute and chronic toxicity of hydrocarbon fractions to aquatic organisms (see Section 3.0).

2.4 Toxicological Data

Toxicological data were compiled from USEPA's ECOTOX database (<http://www.epa.gov/ecotox/>) on September 14, 2005 using search criteria for aquatic animals and specific chemical names. The list of chemical names that was searched in the database is presented in Appendix A (Table A-1). The data field descriptions from ECOTOX are listed and defined in Appendix A (Table A-2).

The downloaded data consisted of acute, chronic, and sublethal effects data from a variety of field and laboratory studies. Laboratory toxicity studies of 24 hours or more were selected over field studies because of the variability and higher potential of confounding factors in the field. Selected toxicity studies included static, flow-through, leaching, and renewal exposures, which were selected over studies listed as dietary exposure or injection. Because there is little to no difference in the sensitivity of freshwater and marine organisms to hydrocarbons, both freshwater and saltwater data were used for invertebrates (e.g., *Daphnia* sp., brine shrimp) and fish (e.g., bluegill, rainbow trout, salmon) (Di Toro et al., 1991; European Community [EC], 2003).

To ensure the robustness of the dataset, both acute and chronic toxicity data were used to derive benchmarks, as recommended by the Technical Guidance Document (TGD) (EC, 2003). Acute data is expressed as the concentration of a toxin (in this case petroleum hydrocarbons) which causes death in 50 percent of the organisms (LC_{50}), or the median sublethal effects concentration (EC_{50}) following exposure for 24 to 96 hours. Chronic toxicological data are expressed as the no observed effects concentration (NOEC) and lowest observed effects concentration (LOEC) following exposure of a particularly sensitive life stage or for a significant fraction of the life span of the test organism.

Acute data were selected using LC_{50} s (also expressed as EC_{50} s, ED_{50} s, or LT_{50} s) and mortality as the endpoint. Results of acute toxicity tests lasting 24 hours or more were used. The short (24 hour) tests are used most frequently for small, short-lived animals, such as water fleas (*Daphnia* spp.) and are equivalent to longer (96 hour or more) acute tests with larger, longer-lived animals, such as fish. Smaller animals are often more sensitive to chemicals than larger animals, so the use of acute data consisting of toxicity tests of various durations did not bias the geometric mean toxicity values for hydrocarbons with different amounts of toxicity data. Chronic data were selected using the NOEC and LOEC for endpoints of growth and reproduction.

Any toxicity data listed as less than (<) were rejected. In addition, three obvious data outliers were removed from the aromatic dataset; there were no obvious data outliers in the aliphatic dataset.

Information such as molecular weight, solubility, and K_{ow} were obtained for each hydrocarbon. The solubility of each hydrocarbon was compared to the toxicity data and any toxicity values that

were substantially higher than the solubility were eliminated, because this indicated the presence of the chemical in a separate phase and, in these cases, mortality occurred for some reasons other than toxicity of the dissolved chemical (*e.g.*, physical abrasion or adherence).

For the aliphatic hydrocarbons, 58 data points were retained, representing a total of 12 hydrocarbons; no chronic data were found. For the aromatic hydrocarbons, 718 acute data points were retained, representing a total of 25 hydrocarbons. Thirty-six chronic data points (18 NOECs and 18 LOECs) were retained for five aromatic hydrocarbons (benzene, xylene, ethylbenzene, fluoranthene, and phenanthrene) and included in the dataset. Data from the ECOTOX downloads for both aliphatic and aromatic hydrocarbons are presented in Appendix A.

3.0 DERIVATION OF RISK-BASED SEDIMENT BENCHMARKS

3.1 Derivation of the Aquatic Toxicity Value

Because the toxicity data were not normally distributed, the geometric mean toxicity value for each hydrocarbon best characterizes the central tendency for the dataset and was used in the regression, as recommended by Hansen et al. (2003). The geometric mean of the acute and chronic toxicity data in mg/L (see Section 2.3) for each hydrocarbon was obtained and converted to millimoles per liter (mM/L) by dividing by its molecular weight. Tables 2 and 3 present the data used to plot the regression equations, including various aliphatic and aromatic hydrocarbons from ECOTOX, their geometric mean LC₅₀s, Log K_{ow}s, and solubility. The log of the mean LC₅₀ was plotted against the Log K_{ow} for both the aromatic and aliphatic hydrocarbons (Figures 4 and 5, respectively) to obtain a relationship between K_{ow} and toxicity. The resulting regression equations were as follows:

$$\text{Aromatic Hydrocarbons: } \text{Log LC}_{50} \text{ (mM/L)} = -1.4347\text{Log K}_{ow} + 3.3624, r^2 = 0.89 \quad (\text{Equation 1})$$

$$\text{Aliphatic Hydrocarbons: } \text{Log LC}_{50} \text{ (mM/L)} = -0.08953\text{Log K}_{ow} + 2.241, r^2 = 0.90 \quad (\text{Equation 2})$$

These equations were used to derive a mean aquatic toxicity value (LC₅₀) for each hydrocarbon fraction, based on the Log K_{ow} of various individual hydrocarbons within the fraction. Results are presented in Tables 4 and 5 and in more detail in Appendix B.

Table 2. Solubility, Mean Log LC₅₀s, and Log K_{ow}s for Aliphatic Hydrocarbons Used to Form the Regression Equation

Chemical Name	Solubility (mg/L) at 25°C ^a	Geometric Mean LC ₅₀ ^b (mg/L)	Log LC ₅₀ ^c (mM/L)	Log K _{ow} ^a
Cyclopentane	156	19.64	-0.55	3.00
Cyclohexane	55	17.20	-0.69	3.44
Pentane	38	11.90	-0.78	3.45
Methylcyclohexane	14	4.46	-1.34	3.88
1,3-Dimethylcyclohexane	11.7	7.71	-1.16	4.01
1,1-Dimethylcyclohexane	10.9	5.99	-1.27	4.05
Dimethylcyclohexane	10.9	2.19	-1.71	4.05
Hexane	9.5	2.97	-1.46	4.11
1,2-Dimethylcyclohexane	6	1.69	-1.82	4.39
1,4-Dimethylcyclohexane	3.84	2.05	-1.74	4.39
Ethylcyclohexane	6.3	3.39	-1.52	4.56
Octane	0.66	0.40	-2.46	5.15

^a. Solubility and Log K_{ow}s from TPHCWG, 1997.

^b. Geometric mean LC₅₀s based on data from ECOTOX database (Appendix A).

^c. Each mean LC₅₀ was then converted to mM/L by dividing by the molecular weight and then obtaining the logarithm at base 10.

Table 3. Solubility, Mean Log LC₅₀s, and Log K_{ow}s for Aromatic Hydrocarbons Used to Form the Regression Equation

Chemical Name	Solubility (mg/L) at 25°C ^a	Geometric Mean LC ₅₀ ^b (mg/L)	Log LC ₅₀ ^c (mM/L)	Log K _{ow} ^a
Benzene	1770	47.37	-0.22	2.13
Toluene	526	27.32	-0.53	2.69
1,2-Dimethylbenzene	178	10.95	-0.99	3.12
Ethylbenzene	169	16.77	-0.80	3.13
1,4-Dimethylbenzene	162	2.90	-1.56	3.15
Xylene	106	13.63	-1.37	3.16
1,3-Dimethylbenzene	161	8.57	-1.09	3.2
Naphthalene	31	3.17	-1.61	3.37
1,2,4-Trimethylbenzene	57	5.28	-1.36	3.58
1,3,5-Trimethylbenzene	48.2	6.87	-1.24	3.58
Isopropylbenzene	61.3	0.601	-2.30	3.63
Propylbenzene	52.2	1.55	-1.89	3.69
2-Methylnaphthalene	24.6	1.48	-1.98	3.86
1-Methylnaphthalene	25.8	4.18	-1.53	3.87
1,1'-Biphenyl	6.94	2.36	-1.82	3.69
2,6-Dimethylnaphthalene	2	0.77	-2.31	4.1
1,2,4,5-Tetramethylbenzene	3.48	0.47	-2.46	4.1
Phenanthrene	1.15	0.11	-3.21	4.36
Dibenzothiophene	1.47	0.28	-2.82	4.38
1,3-Dimethylnaphthalene	8	1.14	-2.14	4.42
Anthracene	0.0434	0.01	-4.37	4.54
Pyrene	0.135	0.003	-4.78	5.18
Fluoranthene	0.26	0.023	-3.95	5.22
Chrysene	0.002	0.0007	-5.51	5.79
Benzo(a)pyrene	0.00162	0.0053	-4.68	6.04

^a. Solubility and Log K_{ow}s from TPHCWG, 1997.

^b. Geometric mean LC₅₀s based on data from ECOTOX database (Appendix A).

^c. Each mean LC₅₀ was then converted to mM/L by dividing by the molecular weight and then obtaining the logarithm at base 10.

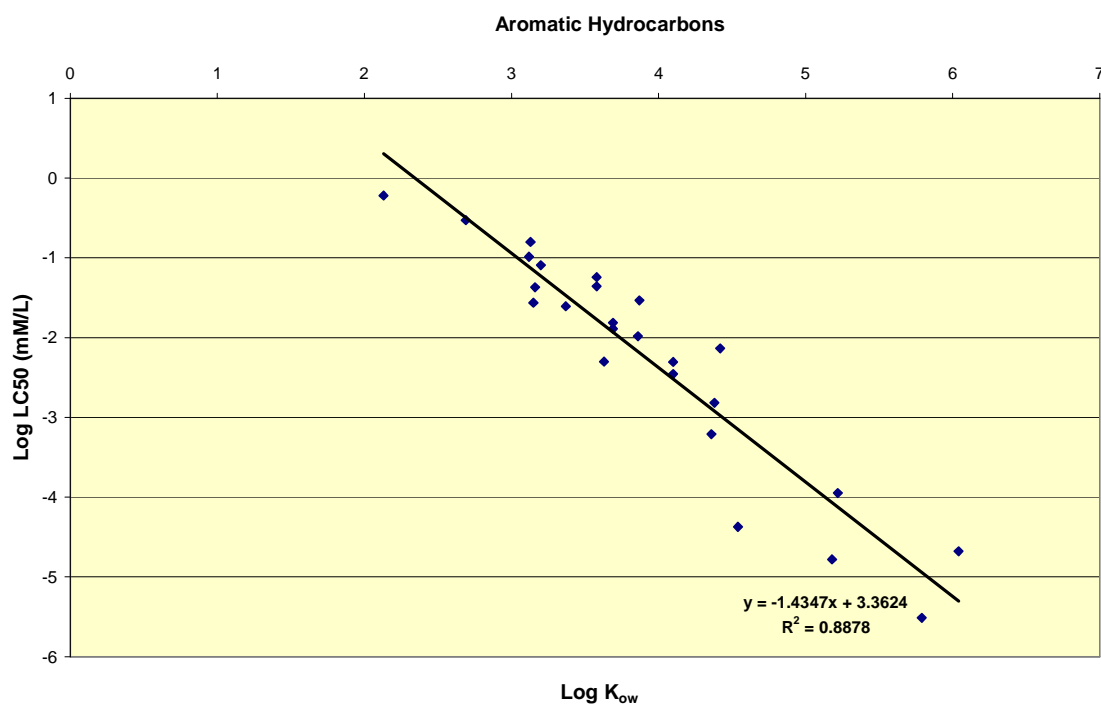


Figure 4. Regression of Log LC₅₀ verses Log K_{ow} for Aromatic Hydrocarbons

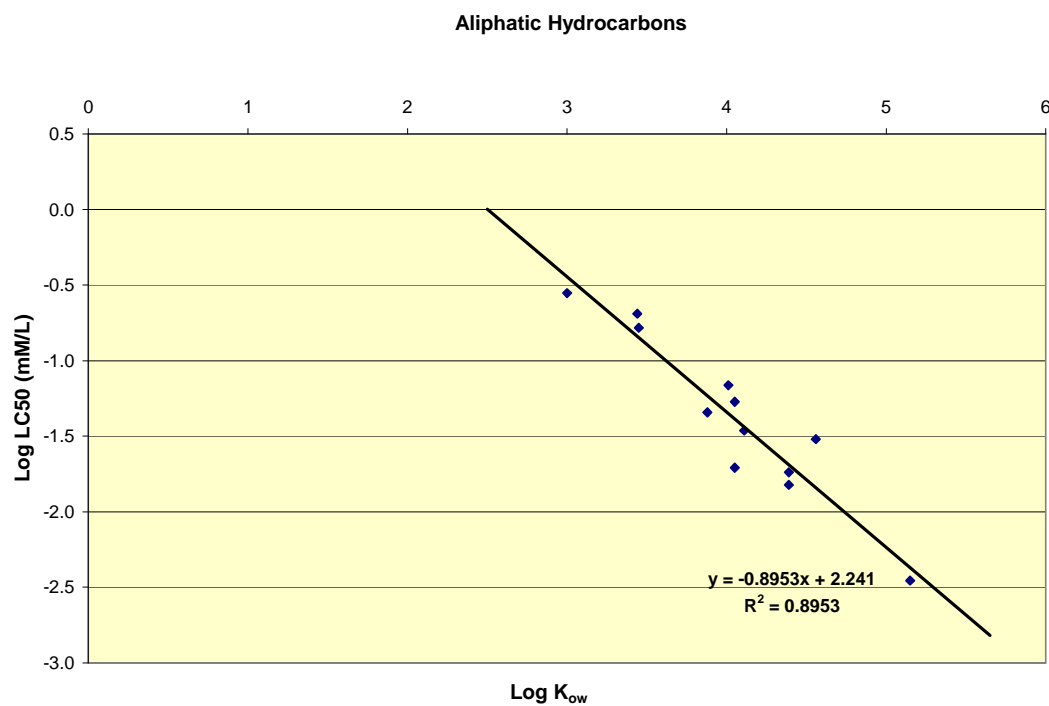


Figure 5. Regression of Log LC₅₀ verses Log K_{ow} for Aliphatic Hydrocarbons

Table 4. Geometric Mean Acute Aquatic Toxicity (LC₅₀) Values for Four Aliphatic Hydrocarbon Fractions

Fraction	Aliphatic Hydrocarbons	Log K _{ow}	Geometric Mean Log K _{ow}	Log LC ₅₀ ^a (mM/L)	LC ₅₀ (mg/L)	Geometric Mean LC ₅₀ ^b (mg/L)
C ₅ -C ₈	n-Pentane	3.45	4.12	-0.85	10.24	3.27
	2,2-Dimethylbutane	3.82		-1.18	5.71	
	Cyclopentane	3.00		-0.44	25.18	
	2,3-Dimethylbutane	3.85		-1.21	5.36	
	2-Methylpentane	3.74		-1.11	6.73	
	3-Methylpentane	3.6		-0.98	8.98	
	n-Hexane	4.11		-1.44	3.14	
	cycloheptane	4.00		-1.34	4.49	
	2,2-Dimethylpentane	4.14		-1.47	3.43	
	Methylcyclopentane	3.37		-0.78	14.09	
	2,4-Dimethylpentane	4.14		-1.47	3.43	
	2,2,3-Trimethylbutane	4.03		-1.37	4.30	
	3,3-Dimethylpentane	4.14		-1.47	3.43	
	Cyclohexane	3.44		-0.84	12.20	
	2,3-Dimethylpentane	4.14		-1.47	3.43	
	3-Methylhexane	4.27		-1.58	2.62	
	2,2,4-Trimethylpentane	4.54		-1.82	1.71	
	Heptane	5.00		-2.24	0.58	
	Methylcyclohexane	3.88		-1.23	5.75	
	1,1,3-Trimethylcyclopentane	4.35		-1.65	2.49	
	2,3,4-Trimethylpentane	4.54		-1.82	1.71	
	2-Methylheptane	4.80		-2.06	1.00	
	3-Methylheptane	4.80		-2.06	1.00	
	1,4-Dimethylcyclohexane	4.39		-1.69	2.29	
	2,2,5-Trimethylhexane	5.06		-2.29	0.66	
	1,2-Dimethylcyclohexane	4.39		-1.69	2.29	
	Octane	5.15		-2.37	0.49	
C ₉ -C ₁₂	3-Methyloctane	5.32	6.01	-2.52	0.39	0.094
	Nonane	5.65		-2.82	0.20	
	Decane	6.25		-3.35	0.06	
	Undecane	6.94		-3.97	0.02	
C ₁₃ -C ₁₈	Tetradecane	7.20	8.57	-4.21	0.0124	0.0008
	Pentadecane	8.63		-5.49	0.0007	
	Hexadecane	8.25		-5.15	0.0016	
	Heptadecane	9.69		-6.43	0.0001	
	Octadecane	9.32		-6.10	0.0002	
C ₁₉ -C ₃₆	Nonadecane	10.72	11.64	-7.36	1.18E-05	1.77E-06
	Eicosane	11.27		-7.85	4.00E-06	
	Tetracosane	13.07		-9.46	1.17E-07	

a Values obtained from the regression equation and corresponding Log K_{ow}.

b Presented as the geometric mean acute aquatic toxicity value. To obtain the final chronic value, the mean LC₅₀ for each hydrocarbon fraction is divided by an application factor (AF) of 15 (Section 3.2).

Table 5. Geometric Mean Acute Aquatic Toxicity (LC₅₀) Values for Four Aromatic Hydrocarbon Fractions

Fraction	Aromatic Hydrocarbons	Log K _{ow}	Geometric Mean Log K _{ow}	Log LC ₅₀ ^a (mM/L)	LC ₅₀ (mg/L)	Geometric Mean LC ₅₀ ^b (mg/L)
C ₆ -C ₈	Benzene	2.13	2.82	0.31	158.18	17.86
	Toluene	2.69		-0.50	29.33	
	Ethylbenzene	3.13		-1.13	7.91	
	p-Xylene	3.18		-1.20	6.70	
	o-Xylene	3.15		-1.16	7.40	
C ₉ -C ₁₂	Isopropylbenzene	3.63	3.94	-1.85	1.72	0.69
	n-Propylbenzene	3.69		-1.93	1.41	
	1-Methyl-4-ethylbenzene	3.63		-1.85	1.72	
	1,3,5-Trimethylbenzene	3.58		-1.77	2.02	
	Naphthalene	3.37		-1.47	4.32	
	2-Methylnaphthalene	3.86		-2.18	0.95	
	1-Methylnaphthalene	3.87		-2.19	0.92	
	2-Ethylnaphthalene	4.40		-2.95	0.18	
	Biphenyl	3.69		-1.93	1.80	
	2,6-Dimethylnaphthalene	4.10		-2.52	0.47	
	2,3-Dimethylnaphthalene	4.40		-2.95	0.18	
	Dibenzothiophene	4.38		-2.92	0.22	
	Acenaphthene	3.92		-2.26	0.84	
	1-Methyl-4-isopropylbenzene	4.10		-2.52	0.41	
	Isobutylbenzene	4.01		-2.39	0.55	
	n-Butylbenzene	4.26		-2.75	0.24	
	Sec-Butylbenzene	4.10		-2.52	0.41	
	1,2,4,5-Tetramethylbenzene	4.10		-2.52	0.41	
C ₁₃ -C ₁₅	Fluorene	3.97	4.67	-2.33	0.77	0.078
	Phenanthrene	4.36		-2.89	0.23	
	2-Methylantracene	5.15		-4.03	0.02	
	9-Methylantracene	5.07		-3.91	0.02	
	1-Methylphenanthrene	4.93		-3.71	0.04	
C ₁₆ -C ₂₄	Benzo(a)fluorene	5.40	5.90	-4.38	0.0089	0.002
	Benz(a)anthracene	5.91		-5.12	0.0017	
	Chrysene	5.79		-4.94	0.0026	
	Benzo(a)pyrene	6.04		-5.30	0.0013	
	Pyrene	5.18		-4.07	0.02	
	Fluoranthene	5.22		-4.13	0.02	
	Coronene	6.75		-6.32	0.0001	
	Benzo(ghi)perylene	6.29		-5.66	0.0006	
	Dibenz(a,h)anthracene	6.75		-6.32	0.0001	

a Values obtained from the regression equation and corresponding Log K_{ow}.

b Presented as the geometric mean acute aquatic toxicity value. To obtain the final chronic value, the mean LC₅₀ for each hydrocarbon fraction is divided by an application factor (AF) of 15 (Section 3.2).

3.2 Derivation of the Application Factor (AF)

Because of the small number of toxicity tests available and the small number of taxa of animals used in the tests, an application factor (AF) was applied to the estimated mean toxicity values to obtain a conservative final chronic toxicity value, which is protective of most aquatic species. The EC (2003), in its TGD, recommends applying an application factor to combine acute and chronic toxicity data and uncertainty about the sensitivity of untested species for a chemical. The application factor recommended by the TGD varies from 10 to 1000, depending on the number of acute and chronic tests performed, and the number of different aquatic taxa tested.

Due to variability between acute and chronic tests, the European Center for Ecotoxicology and Toxicology of Chemicals (ECETOC) evaluated the acute-chronic ratio (ACR) for organic chemicals in aquatic organisms (Länge et al., 1998). The ratios between the median lethal concentration (LC₅₀) or median effects concentration (EC₅₀) and the no observed effects concentrations (NOEC) were determined for 71 chemicals. Länge et al. (1998) calculated the upper 90th percentile ratio for the whole data set and for different subsets of chemicals of different types. The 90th percentile ACR for the entire data set was 73. Highest ACR values were for metals and organochlorine pesticides. A subgroup of chemicals classified as aromatic hydrocarbons had a mean ACR for invertebrates and fish of 6.2 and a 90th percentile ACR of 21.3.

Hansen et al. (2003) estimated a final acute-chronic ratio (FACR) of 4.16 from the geometric mean of various species' mean ACRs for six PAHs. Included in the dataset were toxicity data for the water flea *Daphnia* (2.41), the midge *Paratanytarsus* (6.68), the fathead minnow *Pimephales* (2.61), the rainbow trout *Oncorhynchus* (7.9), the mysid *Americamysis* (3.59), and the killifish *Cyprinodon* (4.36).

ECETOC recommended using an ACR of 15 to 25 in risk assessments for all chemicals except metals and pesticides to adjust for uncertainties in the data between acute and chronic studies. Based on these reported ACRs for hydrocarbons and closely related non-polar organic chemicals and the AF values recommended for non-polar organics in the Technical Guidance Document, the AF used for this report was set at 15. This value represents a conservative estimate of the AF for all hydrocarbons. For example, Suter and Rosen (1988) evaluated the comparative acute and chronic/sublethal toxicity of several chemicals to marine fish and crustaceans. ACRs for aromatic hydrocarbons calculated from their data ranged from 2 to 4. Niederlehner et al. (1998) reported an ACR of 1.76 ± 1.12 for several monocyclic aromatic hydrocarbons and chlorinated ethylenes and the water flea *Ceriodaphnia dubia*. Thus, an AF of 15 is expected to be fully protective of the most sensitive species in the affected freshwater or marine environment.

The AF of 15 was applied to the mean aquatic toxicity value (mean LC₅₀) for each hydrocarbon fraction to account for variability among acute and chronic effects, as well as variability among different species' sensitivities to the various hydrocarbons (Appendix B). The resulting value is a final chronic value (FCV) for each hydrocarbon fraction (Table 6).

3.3 Calculating Sediment Benchmarks

The sediment benchmark for each hydrocarbon fraction was then determined from the final aquatic chronic toxicity value, K_{ow}, K_{oc}, and the fraction of organic carbon in sediment (f_{oc}). The Log organic carbon-to-water partitioning coefficient (Log K_{oc}) is related to Log K_{ow} for each hydrocarbon (Tables 4 and 5) by the following equation (Di Toro et al., 1991):

$$\text{Log } K_{oc} = 0.00028 + 0.938\text{Log } K_{ow} \quad (\text{Equation 3})$$

The K_{oc} for each hydrocarbon fraction was calculated using the above equation and the geometric mean of the Log K_{ow} s (Tables 4 and 5) for the hydrocarbons in the fraction. This was then multiplied by the FCV for each hydrocarbon fraction and by the fraction of organic carbon in the sediment (f_{oc}) (Equation 4). The average amount of organic carbon in site sediments usually ranges between 0.1% and 20%. Nonpolar organic chemicals may adsorb to other solids (*e.g.*, clay particles) in sediments containing less than 0.1 % organic carbon, decreasing the reliability of this EqP approach to estimating sediment benchmarks. An f_{oc} of 0.1% ($f_{oc} = 0.001$) was used here to give the lowest (most conservative) estimated benchmarks.

$$\text{Sediment Benchmark (mg/kg)} = K_{oc} \times \text{FCV} \times f_{oc} (0.001) \quad (\text{Equation 4})$$

Table 6 presents the FCVs and sediment benchmarks (with and without normalization to organic carbon) for each hydrocarbon fraction. The spreadsheet of data and calculations for these equations is included in Appendix B. Sediment benchmarks for aliphatic fractions increase with increasing carbon number in the fractions; sediment benchmarks for aromatic hydrocarbons show the opposite pattern (decreasing values with increasing carbon number). This is caused by the larger increase in K_{ow} and associated K_{oc} with increasing molecular weight (and carbon number) for aliphatic than for aromatic hydrocarbons. Very little of the aliphatic hydrocarbons in the two highest aliphatic carbon fractions (mean log K_{ow} 8.57 and 11.64) desorb from sediment particles into porewater where they are available for bioaccumulation. Thus, it requires more of the high molecular weight aliphatic hydrocarbons in bulk sediments to generate a potentially toxic dose of bioavailable, high molecular weight aliphatic hydrocarbons in sediment porewater.

The FCVs for PAHs in the EPA guidance for deriving ESBs (Hansen et al., 2003) were based on the critical body burden concept (McCarty et al., 1992). Aquatic toxicity values were normalized to a Log K_{ow} of 1 (the Log K_{ow} at which the PAH concentration in water and organism lipids would be equivalent). Resulting ESBs increased with molecular weight from 385 mg/kg oc for naphthalene to 1128 mg/kg oc for dibenz(a,h)anthracene. In the method used in this document, sediment benchmarks were based on non-normalized FCVs based on published or estimated acute toxicity values divided by an application factor. Values decreased with increasing molecular weight (carbon number) from 531 mg/kg oc for the C_6 - C_8 aromatic fraction (mainly BTEX) to 40 for the C_{16} - C_{24} aromatic fraction (mainly four- to six-ring PAHs). This may be due to the lower slope for the K_{oc} versus FCV regression for non-normalized data.

The aqueous solubility of all the aliphatic hydrocarbons in the two highest carbon number aliphatic fractions and the higher molecular weight PAHs in the highest carbon number aromatic fraction is below the estimated acute toxicity value and in many cases the FCV. Thus, the benchmarks for these fractions are conservative. If the benchmarks for these fractions are exceeded, it will be difficult to distinguish between toxicological effects and potential physical impacts. In these cases, consideration should be given to further site evaluations to examine the potential for physical alterations to the habitat, physical impacts to the community structure, or even potential aesthetic impairments. Currently there are no other published sediment guidelines for aliphatic hydrocarbons that can be substituted.

Table 6. Sediment Benchmarks for Recommended Petroleum Hydrocarbon Fractions

Hydrocarbon Fraction	Geometric Mean Log K_{ow}	K_{oc}	Final Chronic Value ($\mu\text{g/L}$)	Sediment Benchmark (mg/kg oc)	Sediment Benchmark ($f_{oc} = 0.001$) (mg/kg)
<i>Aliphatic Hydrocarbons</i>					
$C_5 - C_8$	4.12	7.24×10^3	218	1591	1.59
$C_9 - C_{12}$	6.01	4.37×10^5	6.3	2722	2.72
$C_{13} - C_{18}$	8.57	1.10×10^8	0.05 ^a	5543	5.54
$C_{19} - C_{36}$	11.64	8.32×10^{10}	0.0001 ^a	9883	9.88
<i>Aromatic Hydrocarbons</i>					
$C_6 - C_8$	2.82	4.47×10^2	1191	531	0.53
$C_9 - C_{12}$	3.94	4.90×10^3	46.2	228	0.23
$C_{13} - C_{15}$	4.67	2.40×10^4	5.2	125	0.13
$C_{16} - C_{24}$	5.9	3.39×10^5	0.12 ^a	40	0.04

^a The fraction is not likely toxic because mean LC_{50} exceeds mean aqueous solubility.

3.4 Application of Sediment Benchmarks

The sediment benchmark values that are normalized to 0.1 percent organic carbon can be applied to any petroleum contaminated sediment site. However, it should be noted that a 0.1 percent organic carbon content in sediments represents the low range of organic carbon, resulting in the most stringent benchmarks. The equilibrium partitioning approach is appropriate for sediments containing 0.1 percent or more of organic matter (Di Toro et al., 1991). When site-specific organic carbon concentrations are available, it is recommended that an average site-specific organic carbon content be used to derive a site-specific criterion. This can be accomplished by inserting the appropriate fraction of organic carbon (f_{oc}) into Equation 4 to obtain a site-specific sediment benchmark for each hydrocarbon fraction. The benchmarks would then be compared to the analytical chemistry results for each hydrocarbon fraction and screened to determine the potential for site risk to aquatic organisms.

As discussed in Section 1.2.3.2, the current MADEP methods use different hydrocarbon fractions that span a wide range of carbon numbers and physical/chemical and toxicological properties. These fractions are currently used for the estimation of risk to human health from petroleum hydrocarbons. Sediment benchmarks were also developed for these fractions as a comparison and are listed in Table 7. The detailed calculation spreadsheet is available in Appendix B (Table B-2).

Table 7. Sediment Benchmarks for Current MADEP Petroleum Hydrocarbon Fractions

Hydrocarbon Fraction	Geometric Mean Log K_{ow}	K_{oc}	Final Chronic Value ($\mu\text{g/L}$)	Sediment Benchmark (mg/kg oc)	Sediment Benchmark ($f_{oc} = 0.001$) (mg/kg)
<i>Aliphatic Hydrocarbons</i>					
$C_5 - C_8$	4.12	7.24×10^3	218	1591	1.59
$C_9 - C_{18}$	7.32	7.41×10^6	0.4	3167	3.17
$C_{19} - C_{36}$	11.64	8.32×10^{10}	0.0001 ^a	9883	9.88
<i>Aromatic Hydrocarbons</i>					
$C_9 - C_{10}$	3.84	3.98×10^3	59.4	236	0.24
$C_{11} - C_{22}$	4.81	3.31×10^4	2.8	92	0.09

^a The fraction is not likely toxic because mean LC_{50} exceeds mean aqueous solubility.

3.5 Uncertainties

There are several uncertainties inherent in the use of the equilibrium partitioning method for developing sediment benchmarks for petroleum hydrocarbon fractions, which are accounted for in the method: varying values for Log K_{ow} , the appropriate value for the application factor, the limited amount of toxicity data for aliphatic hydrocarbons, the use of toxicity data for both marine and freshwater species, and the use of acute and chronic toxicity data for tests performed over various lengths of time. Each of these uncertainties is further discussed below.

Log K_{ow}

Several methods have been used over the years to measure or model Log K_{ow} for non-polar organic chemicals, including hydrocarbons. In general, water, octanol, and the chemical of concern are mixed and allowed to equilibrate, allowed to settle, and then the concentration of chemical is measured in both phases. Originally, a shake-flask method was used, until it was discovered that the approach left enough octanol in the water to act as a detergent. Nowadays, slow stir is the method of choice. Gas chromatography is the most common analytical method (Renner, 2002). Other methods include estimating the K_{ow} by the linear relationship between log aqueous solubility and log K_{ow} or by modeling K_{ow} by structure/activity relationships for molecular structure, molecular volume, and various other physical and chemical properties. The different estimation methods produce values that may vary by an order of magnitude or more, particularly for high molecular weight compounds with extremely low aqueous solubilities (Güsten et al., 1991; Mackay et al., 1992; de Maagd et al., 1998). The most representative, most widely used values for Log K_{ow} , as recommended by Mackay et al. (1992) in TPHCWG (1997), were used in this report.

Application Factor

Because of the small number of toxicity tests available and the small number of taxa of animals used in the tests, an AF of 15 was applied to the estimated mean toxicity values to obtain a conservative final chronic toxicity value, which is protective of most aquatic species. The uncertainty in the geometric mean toxicity values for individual hydrocarbons from the ECOTOX database depends on the number of acute and chronic toxicity tests performed and the number of different major animal taxa (e.g., crustaceans, annelids, mollusks, fish) tested (EC, 2003). If few, mainly acute, tests were performed on just a few taxa, the resulting mean toxicity value is less certain than mean values based on a large number of acute and chronic tests with several

species representing a wide taxonomic range. There is a paucity of good aquatic toxicity data for higher molecular weight aromatic hydrocarbons and aliphatic hydrocarbons, mainly because of difficulty in getting enough of these low-solubility chemicals into solution to cause biological effects. Thus, high AFs are needed to be sufficiently protective of more sensitive species and life stages, which is the approach taken in this report. To ensure adequate protection of multiple species, the selected AF (15) is higher than that recommended by ECETOC (Länge et al., 1998) and Hansen et al. (2003).

Limited Toxicity Data for Aliphatic Hydrocarbons

Toxicity data for aliphatic hydrocarbons were limited, especially for the higher molecular weight fractions. The reason for this lack of data is that the solubility of these hydrocarbons is so low, that it is difficult to get enough hydrocarbons into solution to cause chemical toxicity. As a major fraction of most petroleum products, these hydrocarbons contribute to the physical effects of petroleum contamination in sediments by changing the sediment's texture, porosity, oxygen levels, and redox potential, all of which can cause adverse effects to aquatic and benthic organisms. It is difficult, however, to develop sediment benchmarks for these physical and aesthetic effects in sediments.

Marine vs. Freshwater Species

Both marine and freshwater toxicity data were used in this approach to develop sediment benchmarks for petroleum because it is important to use the largest amount of available, reliable aquatic toxicity data to develop appropriate benchmarks. Acute and chronic toxicity data for freshwater and marine species, spanning a wide taxonomic range, are used to derive the most representative mean toxicity values for each hydrocarbon (Di Toro et al., 1991, 2000, 2007; Hansen et al., 2003). Because bioaccumulation of non-polar organic compounds (such as hydrocarbons) is only moderately affected by ambient water salinity, toxicity of these chemicals to freshwater and marine organisms is similar. Sediment benchmarks based on both freshwater and marine data are equally applicable and protective in both marine and freshwater ecosystems.

Toxicity Test Duration

The duration of aquatic toxicity tests used to derive sediment benchmarks ranged from 24 hours to months (life cycle tests are often performed for a significant fraction of the normal life-span of the test organism). The 24 hour tests are usually performed on small, short-lived organisms, such as water fleas, and brine shrimp. These organisms equilibrate rapidly with dissolved chemicals in the ambient water and therefore respond to toxicants more quickly than do larger animals. Inclusion of long term tests in the toxicity dataset is a conservative strategy intended, along with the acute-chronic ratio, to protect more sensitive, untested species. This approach was used by Di Toro et al. (1991, 2000), Hansen et al. (2003), and the EC (2003) to develop benchmarks for non-polar organic chemicals in water and sediments.

Overall, these uncertainties were addressed in an appropriately conservative manner consistent with previous work by Hansen et al. (2003), DiToro et al. (1991), Total Petroleum Hydrocarbon Criteria Working Group (TPHCWG) (1997), and the European Community (EC) (2003).

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Appendix A
Toxicological Data

Table A-1. Various Hydrocarbon Names Used to Search ECOTOX Database

Aliphatic Hydrocarbons	Aromatic Hydrocarbons
n-Pentane	Benzene
2,2-Dimethylbutane	Toluene
Cyclopentane	Ethylbenzene
2,3-Dimethylbutane	p-Xylene
2-Methylpentane	o-Xylene
3-Methylpentane	Isopropylbenzene
n-Hexane	n-Propylbenzene
2,2-Dimethylpentane	1-Methyl-4-ethylbenzene
Methylcyclopentane	1,3,5-Trimethylbenzene
2,4-Dimethylpentane	Isobutylbenzene
2,2,3-Trimethylbutane	Sec-Butylbenzene
3,3-Dimethylpentane	1-Methyl-4-isopropylbenzene
Cyclohexane	n-Butylbenzene
2-Methylhexane	1,2,4,5-Tetramethylbenzene
2,3-Dimethylpentane	Naphthalene
3-Methylhexane	2-Methylnaphthalene
2,2,4-Trimethylpentane	1-Methylnaphthalene
Heptane	2-Ethylnaphthalene
n-Propylcyclopentane	Biphenyl
Methylcyclohexane	2,6-Dimethylnaphthalene
1,1,3-Trimethylcyclopentane	2,3-Dimethylnaphthalene
2,3,4-Trimethylpentane	Acenaphthene
2,3-Dimethylhexane	Fluorene
2-Methylheptane	Phenanthrene
3-Methylheptane	Anthracene
1,4-Dimethylcyclohexane	1-Methylphenanthrene
2,2,5-Trimethylhexane	Pyrene
1,2-Dimethylcyclohexane	Fluoranthene
Octane	Benzo(a)fluorene
4-Methyloctane	Benz(a)anthracene
3-Methyloctane	Chrysene
Nonane	Benzo(a)pyrene
Decane	Coronene
Undecane	Benzo(ghi)perylene
Dodecane	Dibenz(a,h)anthracene
Tetradecane	
Pentadecane	
Hexadecane	
Heptadecane	
Octadecane	
Nonadecane	
Eicosane	
Tetracosane	

Table A-2. Field Code Descriptions for ECOTOX Data

Field Code	Description	Manipulations/Selected Data
Test Loc	Test Location = lab or field	Only lab studies were retained
CAS #	Chemical Abstracts Service Number	Retained
Chemical Name	CAS Collective Index Name	Retained
Scientific Name	Species scientific name	Retained
Common Name	Species common name	Retained
Endpoint	The study endpoint - refer to ECOTOX for the full list.	Only LC ₅₀ , EC ₅₀ , ED ₅₀ , LT ₅₀ , NOEC, LOEC were retained.
Effect	The study effects - refer to ECOTOX for the full list.	For acute data, only mortality was selected; for chronic data survival and reproduction were selected.
Effect Measurement	Same as effect	For acute data, only mortality was selected; for chronic data survival and reproduction were selected.
Trend	The observed or measured response trend as compared to the control is coded when textually or graphically reported.	All retained
Media Type	FW = freshwater; SW = saltwater	Both retained
Test Duration	Length of study	All studies greater than 24 hours were retained.
Duration Units	Days or hours	All converted to hours if not reported as such.
Exposure Type	F = flow through; S = static; R = renewal; L = leachate; I = Injection; D = Diet	I and D rejected; all others retained
Significant	Statistical analysis as compared to the controls in the test result.	All retained
Level	The level of significance, if reported.	All retained
Response Site	A response site or tissue code is used to identify specific body, organ or tissue effect sites for associated effect measurement.	All retained
BCF1 Mean Op	Bioconcentration factor operation – not applicable	Column not used
BCF1 Mean	Bioconcentration factor – not applicable	Column not used
BCF1 Min Op	Bioconcentration factor operation – not applicable	Column not used
BCF1 Min	Bioconcentration factor – not applicable	Column not used
BCF1 Max Op	Bioconcentration factor operation – not applicable	Column not used
BCF1 Max	Bioconcentration factor – not applicable	Column not used

Table A-2. Field Code Descriptions for ECOTOX Data (continued)

Field Code	Description	Manipulations/Selected Data
BCF2 Mean Op	Bioconcentration factor operation – not applicable	Column not used
BCF2 Mean	Bioconcentration factor – not applicable	Column not used
BCF2 Min Op	Bioconcentration factor operation – not applicable	Column not used
BCF2 Min	Bioconcentration factor – not applicable	Column not used
BCF2 Max Op	Bioconcentration factor operation – not applicable	Column not used
BCF2 Max	Bioconcentration factor – not applicable	Column not used
Author	Author of study	Retained
Year	Year published	Retained
Title	Article Title	Retained
Ref Source	Journal Title	Retained
Concentration1 Mean Op	Operation signs such as <, >, =	Data associated with <, > values not retained; data associated with = were retained.
Concentration1 Mean	Mean reported concentration	Retained
Concentration1 Min Op	Operation signs such as <, >, =	Data associated with <, > values not retained; data associated with = were retained.
Concentration1 Min	Minimum reported concentration	If the mean and max concentrations were the only two fields reported, the mean of the two was taken.
Concentration1 Max Op	Operation signs such as <, >, =	Data associated with <, > values not retained; data associated with = were retained.
Concentration1 Max	Maximum reported concentration	If the mean and max concentrations were the only two fields reported, the mean of the two was taken.
Concentration1 Type	Concentrations based on the active ingredient (A) or formulation (F)	Both retained
Concentration2 Mean Op	Not reported (NR)	Column not used
Concentration2 Mean	Not reported (NR)	Column not used
Concentration2 Min Op	Not reported (NR)	Column not used
Concentration2 Min	Not reported (NR)	Column not used
Concentration2 Max Op	Not reported (NR)	Column not used
Concentration2 Max	Not reported (NR)	Column not used
Concentration2 Type	Not reported (NR)	Column not used
Concentration Units	Concentration units	All converted to µg/L if not reported as such.
Application Rate	Not applicable	Column not used
Application Units	Not applicable	Column not used
Application Type	Not applicable	Column not used
Application Frequency	Not applicable	Column not used

Table A-2. Field Code Descriptions for ECOTOX Data (continued)

Field Code	Description	Manipulations/Selected Data
Application Date	Not applicable	Column not used
Application Season	Not applicable	Column not used

Table A-3. Aromatic Hydrocarbon Toxicity Data (pages A-5 through A-27)

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
[1,1'-Biphenyl]-4-ol	<i>Daphnia magna</i>	Water flea	48	LC50	3660	Carlson and Caple, 1977
1,1'-Biphenyl	<i>Artemia salina</i>	Brine shrimp	24	LC50	4009	Abernethy et al., 1986
1,1'-Biphenyl	<i>Artemia salina</i>	Brine shrimp	24	LC50	4009	Abernethy et al., 1986
1,1'-Biphenyl	<i>Cyprinodon variegatus</i>	Sheepshead minnow	96	LC50	4600	Dill et al., 1982
1,1'-Biphenyl	<i>Daphnia magna</i>	Water flea	24	LC50	1300	Gersich et al., 1989
1,1'-Biphenyl	<i>Daphnia magna</i>	Water flea	48	LC50	4700	LeBlanc, 1980
1,1'-Biphenyl	<i>Daphnia magna</i>	Water flea	48	LC50	2100	Dill et al., 1982
1,1'-Biphenyl	<i>Daphnia magna</i>	Water flea	48	LC50	360	Gersich et al., 1989
1,1'-Biphenyl	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	4700	Dill et al., 1982
1,1'-Biphenyl	<i>Oncorhynchus mykiss</i>	Rainbow trout, donaldson trout	96	LC50	1500	Dill et al., 1982
1,1'-Biphenyl	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	1950	Brooke, 1991
1,1'-Biphenyl	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	3500	Brooke, 1991
1,1'-Biphenyl	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	2940	Brooke, 1991
1,1'-Biphenyl	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	1450	Brooke, 1991
1,2,4,5-Tetramethylbenzene	<i>Daphnia magna</i>	Water flea	48	LC50	470	Abernethy et al., 1986
1,2,4-Trimethylbenzene	<i>Daphnia magna</i>	Water flea	48	LC50	3606	Abernethy et al., 1986
1,2,4-Trimethylbenzene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	7720	Geiger et al., 1986
1,2-Dimethylbenzene	<i>Morone saxatilis</i>	Striped bass	96	LC50	9642	Benville and Korn, 1977
1,2-Dimethylbenzene	<i>Oncorhynchus mykiss</i>	Rainbow trout	96	LC50	7600	Galassi et al., 1988
1,2-Dimethylbenzene	<i>Poecilia reticulata</i>	Guppy	96	LC50	12000	Galassi et al., 1988
1,2-Dimethylbenzene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	16400	Geiger et al., 1990
1,2-Dimethylbenzene	<i>Oncorhynchus mykiss</i>	Rainbow trout	96	LC50	8050	Holcombe et al., 1987
1,2-Dimethylbenzene	<i>Carassius auratus</i>	Goldfish	96	LC50	16100	Holcombe et al., 1987
1,2-Dimethylbenzene	<i>Catostomus commersoni</i>	White sucker	96	LC50	16100	Holcombe et al., 1987
1,2-Dimethylbenzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	16100	Holcombe et al., 1987
1,2-Dimethylbenzene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	16100	Holcombe et al., 1987
1,2-Dimethylbenzene	<i>Daphnia magna</i>	Water flea	48	LC50	3168	Abernethy et al., 1986
1,3-Dimethylbenzene	<i>Morone saxatilis</i>	Striped bass	96	LC50	8064	Benville and Korn, 1977
1,3,5-Trimethylbenzene	<i>Carassius auratus</i>	Goldfish	96	LC50	12520	Brenniman et al., 1976
1,3,5-Trimethylbenzene	<i>Cancer magister</i>	Dungeness, Edible crab	96	LC50	4300	Caldwell et al., 1977
1,3,5-Trimethylbenzene	<i>Daphnia magna</i>	Water flea	48	LC50	6010	Abernethy et al., 1986
1,3-Dimethylbenzene	<i>Oncorhynchus mykiss</i>	Rainbow trout	96	LC50	8400	Galassi et al., 1988
1,3-Dimethylbenzene	<i>Poecilia reticulata</i>	Guppy	96	LC50	12900	Galassi et al., 1988
1,3-Dimethylbenzene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	16000	Geiger et al., 1990
1,3-Dimethylbenzene	<i>Ceriodaphnia dubia</i>	Water flea	168	LC50	2974	Niederlehner et al., 1998

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
1,3-Dimethylbenzene	<i>Daphnia magna</i>	Water flea	48	LC50	9558	Abernethy et al., 1986
1,3-Dimethylnaphthalene	<i>Oncorhynchus mykiss</i>	Rainbow trout	96	LC50	1700	Edsall, 1991
1,3-Dimethylnaphthalene	<i>Daphnia pulex</i>	Water flea	48	LC50	767	Smith et al., 1988
1,4-Dimethylbenzene	<i>Crangon franciscorum</i>	Bay shrimp	96	LC50	1753	Benville and Korn, 1977
1,4-Dimethylbenzene	<i>Morone saxatilis</i>	Striped bass	96	LC50	1753	Benville and Korn, 1977
1,4-Dimethylbenzene	<i>Oncorhynchus mykiss</i>	Rainbow trout	96	LC50	2600	Galassi et al., 1988
1,4-Dimethylbenzene	<i>Poecilia reticulata</i>	Guppy	96	LC50	8800	Galassi et al., 1988
1-Methylnaphthalene	<i>Artemia salina</i>	Brine shrimp	24	LC50	2560	Abernethy et al., 1986
1-Methylnaphthalene	<i>Cancer magister</i>	Dungeness or edible crab	48	LC50	8200	Caldwell et al., 1977
1-Methylnaphthalene	<i>Cancer magister</i>	Dungeness or edible crab	96	LC50	1900	Caldwell et al., 1977
1-Methylnaphthalene	<i>Cyprinodon variegatus</i>	Sheepshead minnow	24	LC50	3400	Anderson et al., 1974
1-Methylnaphthalene	<i>Nitocra spinipes</i>	Harpacticoid copepod	96	LC50	13000	Bengtsson and Tarkpea, 1983
1-Methylnaphthalene	<i>Pimephales promelas</i>	Fathead minnow	24	LC50	9000	Mattson et al., 1976
1-Methylnaphthalene	<i>Pimephales promelas</i>	Fathead minnow	48	LC50	9000	Mattson et al., 1976
1-Methylnaphthalene	<i>Pimephales promelas</i>	Fathead minnow	72	LC50	9000	Mattson et al., 1976
1-Methylnaphthalene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	9000	Mattson et al., 1976
2,6-Dimethylnaphthalene	<i>Eurytemora affinis</i>	Calanoid copepod	24	LC50	852	Ott et al., 1978
2,6-Dimethylnaphthalene	<i>Palaemonetes pugio</i>	Daggerblade grass shrimp	96	LC50	700	Neff et al., 1976
2-Methylnaphthalene	<i>Artemia salina</i>	Brine shrimp	24	LC50	474	Abernethy et al., 1986
2-Methylnaphthalene	<i>Artemia salina</i>	Brine shrimp	24	LC50	4735	Abernethy et al., 1986
2-Methylnaphthalene	<i>Cancer magister</i>	Dungeness or edible crab	48	LC50	5000	Caldwell et al., 1977
2-Methylnaphthalene	<i>Cancer magister</i>	Dungeness or edible crab	96	LC50	1300	Caldwell et al., 1977
2-Methylnaphthalene	<i>Cyprinodon variegatus</i>	Sheepshead minnow	24	LC50	2000	Anderson et al., 1974
2-Methylnaphthalene	<i>Eurytemora affinis</i>	Calanoid copepod	24	LC50	1499	Ott et al., 1978
2-Methylnaphthalene	<i>Oncorhynchus mykiss</i>	Rainbow trout, donaldson trout	24	LC50	2443	Kennedy, 1990
2-Methylnaphthalene	<i>Oncorhynchus mykiss</i>	Rainbow trout, donaldson trout	48	LC50	2080	Kennedy, 1990
2-Methylnaphthalene	<i>Oncorhynchus mykiss</i>	Rainbow trout, donaldson trout	72	LC50	1694	Kennedy, 1990
2-Methylnaphthalene	<i>Oncorhynchus mykiss</i>	Rainbow trout, donaldson trout	96	LC50	1456	Kennedy, 1990
2-Methylnaphthalene	<i>Palaemonetes pugio</i>	Daggerblade grass shrimp	24	LC50	1700	Anderson et al., 1974
2-Methylnaphthalene	<i>Palaemonetes pugio</i>	Daggerblade grass shrimp	24	LC50	1650	Tatem, 1975
2-Methylnaphthalene	<i>Palaemonetes pugio</i>	Daggerblade grass shrimp	48	LC50	1400	Tatem, 1975
2-Methylnaphthalene	<i>Palaemonetes pugio</i>	Daggerblade grass shrimp	96	LC50	1100	Tatem et al., 1978
2-Methylnaphthalene	<i>Palaemonetes pugio</i>	Daggerblade grass shrimp	96	LC50	1100	Tatem, 1975

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
2-Methylnaphthalene	<i>Palaemonetes pugio</i>	Daggerblade grass shrimp	96	LC50	1100	Neff et al., 1976
2-Methylnaphthalene	<i>Penaeus aztecus</i>	Brown shrimp	24	LC50	700	Anderson et al., 1974
2-Methylnaphthalene	<i>Penaeus aztecus</i>	Brown shrimp	96	LC50	600	Tatem et al., 1978
Anthracene	<i>Americamysis bahia</i>	Opossum shrimp	48	LC50	3.6	Pelletier et al., 1997
Anthracene	<i>Chironomus tentans</i>	Midge	240	LC50	6	Hatch, 1999
Anthracene	<i>Hyalella azteca</i>	Scud	240	LC50	5.6	Hatch, 1999
Anthracene	<i>Lepomis macrochirus</i>	Bluegill	48	LC50	9.69	McCloskey and Oris, 1991
Anthracene	<i>Lepomis macrochirus</i>	Bluegill	48	LC50	3.36	McCloskey and Oris, 1991
Anthracene	<i>Lepomis macrochirus</i>	Bluegill	48	LC50	11.56	McCloskey and Oris, 1991
Anthracene	<i>Lepomis macrochirus</i>	Bluegill	48	LC50	10.05	McCloskey and Oris, 1991
Anthracene	<i>Lepomis macrochirus</i>	Bluegill	48	LC50	5.1	McCloskey and Oris, 1991
Anthracene	<i>Lepomis macrochirus</i>	Bluegill	48	LC50	12.02	McCloskey and Oris, 1991
Anthracene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	2.78	Oris, 1985
Anthracene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	4.5	Oris, 1986
Anthracene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	46	Oris, 1986
Anthracene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	7.47	McCloskey and Oris, 1991
Anthracene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	1.27	McCloskey and Oris, 1991
Anthracene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	7.97	McCloskey and Oris, 1991
Anthracene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	6.78	McCloskey and Oris, 1991
Anthracene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	3.74	McCloskey and Oris, 1991
Anthracene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	8.27	McCloskey and Oris, 1991
Anthracene	<i>Lepomis sp.</i>	Sunfish	96	LC50	26.47	Oris, 1986
Anthracene	<i>Lepomis sp.</i>	Sunfish	96	LC50	18.23	Oris, 1985
Anthracene	<i>Lepomis sp.</i>	Sunfish	96	LC50	11.92	Oris, 1985
Anthracene	<i>Lepomis macrochirus</i>	Bluegill	202	LT50	15	Oris, 1986
Benzene	<i>Ceriodaphnia dubia</i>	Water flea	24	LC50	18400	Marchini et al., 1993
Benzene	<i>Mugil curema</i>	White mullet	48	LC50	22000	Correa and Garcia, 1990
Benzene	<i>Pimephales promelas</i>	Fathead minnow	168	LC50	14010	Marchini et al., 1992
Benzene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	24600	Geiger et al., 1990
Benzene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	12600	Geiger et al., 1990
Benzene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	15590	Marchini et al., 1992
Benzene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	24600	Marchini et al., 1992

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
Benzene	<i>Platichthys flesus</i>	Starry, european flounder	97	LC50	10864	Furay and Smith, 1995
Benzene	<i>Ambystoma mexicanum</i>	Mexican axolotl	24	LC50	440000	Slooff, 1982
Benzene	<i>Ambystoma mexicanum</i>	Mexican axolotl	48	LC50	370000	Slooff and Baerselman, 1980
Benzene	<i>Ambystoma mexicanum</i>	Mexican axolotl	48	LC50	370000	Slooff, 1982
Benzene	<i>Artemia salina</i>	Brine shrimp	24	LC50	127325	Abernethy et al., 1986
Benzene	<i>Artemia sp.</i>	Brine shrimp	48	LC50	97800	MacLean and Doe, 1989
Benzene	<i>Artemia sp.</i>	Brine shrimp	48	LC50	139000	MacLean and Doe, 1989
Benzene	<i>Asellus aquaticus</i>	Aquatic sowbug	48	LC50	120000	Slooff, 1983
Benzene	<i>Chironomus thummi</i>	Midge	48	LC50	100000	Slooff, 1983
Benzene	<i>Corixa punctata</i>	Water boatman	48	LC50	48000	Slooff, 1983
Benzene	<i>Daphnia magna</i>	Water flea	24	LC50	250000	LeBlanc, 1980
Benzene	<i>Daphnia magna</i>	Water flea	48	LC50	200000	LeBlanc, 1980
Benzene	<i>Daphnia magna</i>	Water flea	48	LC50	682000	Eastmond et al., 1984
Benzene	<i>Daphnia magna</i>	Water flea	48	LC50	96200	MacLean and Doe, 1989
Benzene	<i>Daphnia magna</i>	Water flea	48	LC50	99200	MacLean and Doe, 1989
Benzene	<i>Daphnia magna</i>	Water flea	48	LC50	76900	MacLean and Doe, 1989
Benzene	<i>Daphnia magna</i>	Water flea	48	LC50	59600	MacLean and Doe, 1989
Benzene	<i>Daphnia magna</i>	Water flea	48	LC50	156600	MacLean and Doe, 1989
Benzene	<i>Daphnia magna</i>	Water flea	48	LC50	135700	MacLean and Doe, 1989
Benzene	<i>Daphnia pulex</i>	Water flea	96	LC50	15000	Trucco et al., 1983
Benzene	<i>Diaptomus forbesi</i>	Calanoid copepod	96	LC50	710000	Saha and Konar, 1983
Benzene	<i>Gammarus pseudolimnaeus</i>	Scud	96	LC50	12100	Brooke, 1987
Benzene	<i>Gammarus pulex</i>	Scud	48	LC50	42000	Slooff, 1983
Benzene	<i>Hydra oligactis</i>	Hydra	48	LC50	34000	Slooff et al., 1983
Benzene	<i>Hydra oligactis</i>	Hydra	48	LC50	34000	Slooff, 1983
Benzene	<i>Ictalurus punctatus</i>	Channel catfish	24	LC50	425000	Mayer and Ellersieck, 1986
Benzene	<i>Ictalurus punctatus</i>	Channel catfish	96	LC50	425000	Mayer and Ellersieck, 1986
Benzene	<i>Katelysia opima</i>	Marine bivalve	24	LC50	225000	Dange and Masurekar, 1984
Benzene	<i>Katelysia opima</i>	Marine bivalve	48	LC50	205000	Dange and Masurekar, 1984
Benzene	<i>Katelysia opima</i>	Marine bivalve	72	LC50	195000	Dange and Masurekar, 1984
Benzene	<i>Katelysia opima</i>	Marine bivalve	96	LC50	190000	Dange and Masurekar, 1984
Benzene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	400000	Mayer and Ellersieck, 1986

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
Benzene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	140000	Mayer and Ellersieck, 1986
Benzene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	910000	Mayer and Ellersieck, 1986
Benzene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	740000	Mayer and Ellersieck, 1986
Benzene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	580000	Mayer and Ellersieck, 1986
Benzene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	370000	Mayer and Ellersieck, 1986
Benzene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	260000	Mayer and Ellersieck, 1986
Benzene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	102000	Mayer and Ellersieck, 1986
Benzene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	165000	Mayer and Ellersieck, 1986
Benzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	230000	Mayer and Ellersieck, 1986
Benzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	100000	Mayer and Ellersieck, 1986
Benzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	600000	Mayer and Ellersieck, 1986
Benzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	450000	Mayer and Ellersieck, 1986
Benzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	290000	Mayer and Ellersieck, 1986
Benzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	370000	Mayer and Ellersieck, 1986
Benzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	260000	Mayer and Ellersieck, 1986
Benzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	102000	Mayer and Ellersieck, 1986
Benzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	165000	Mayer and Ellersieck, 1986
Benzene	<i>Lymnaea stagnalis</i>	Great pond snail	24	LC50	440000	Slooff, 1982
Benzene	<i>Lymnaea stagnalis</i>	Great pond snail	48	LC50	230000	Slooff et al., 1983
Benzene	<i>Lymnaea stagnalis</i>	Great pond snail	48	LC50	230000	Slooff, 1983
Benzene	<i>Lymnaea stagnalis</i>	Great pond snail	48	LC50	230000	Slooff, 1982
Benzene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	24	LC50	9200	Mayer and Ellersieck, 1986
Benzene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	48	LC50	56000	Slooff et al., 1983
Benzene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	21637	Hodson et al., 1984
Benzene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	5900	Galassi et al., 1988
Benzene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	5300	Degraeve et al., 1982
Benzene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	9200	Mayer and Ellersieck, 1986
Benzene	<i>Oryzias latipes</i>	Medaka, high-eyes	24	LC50	74000	Tsuji et al., 1986
Benzene	<i>Oryzias latipes</i>	Medaka, high-eyes	24	LC50	70000	Tsuji et al., 1986
Benzene	<i>Oryzias latipes</i>	Medaka, high-eyes	24	LC50	54000	Tsuji et al., 1986
Benzene	<i>Oryzias latipes</i>	Medaka, high-eyes	48	LC50	250000	Slooff et al., 1983
Benzene	<i>Oryzias latipes</i>	Medaka, high-eyes	48	LC50	74000	Tsuji et al., 1986

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
Benzene	<i>Oryzias latipes</i>	Medaka, high-eyes	48	LC50	70000	Tsuji et al., 1986
Benzene	<i>Oryzias latipes</i>	Medaka, high-eyes	48	LC50	54000	Tsuji et al., 1986
Benzene	<i>Pimephales promelas</i>	Fathead minnow	24	LC50	78000	Slooff, 1982
Benzene	<i>Pimephales promelas</i>	Fathead minnow	48	LC50	84000	Slooff et al., 1983
Benzene	<i>Pimephales promelas</i>	Fathead minnow	48	LC50	84000	Slooff, 1982
Benzene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	12500	Brooke, 1987
Benzene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	35700	Brooke, 1987
Benzene	<i>Poecilia reticulata</i>	Guppy	96	LC50	28600	Galassi et al., 1988
Benzene	<i>Scylla serrata</i>	Crab	24	LC50	7146	Rao et al., 1988
Benzene	<i>Scylla serrata</i>	Crab	24	LC50	9117	Rao et al., 1988
Benzene	<i>Scylla serrata</i>	Crab	24	LC50	11299	Rao et al., 1988
Benzene	<i>Scylla serrata</i>	Crab	24	LC50	7146	Rao et al., 1988
Benzene	<i>Scylla serrata</i>	Crab	24	LC50	5614	Rao et al., 1988
Benzene	<i>Scylla serrata</i>	Crab	48	LC50	5799	Rao et al., 1988
Benzene	<i>Scylla serrata</i>	Crab	48	LC50	8052	Rao et al., 1988
Benzene	<i>Scylla serrata</i>	Crab	48	LC50	10014	Rao et al., 1988
Benzene	<i>Scylla serrata</i>	Crab	48	LC50	5799	Rao et al., 1988
Benzene	<i>Scylla serrata</i>	Crab	48	LC50	4418	Rao et al., 1988
Benzene	<i>Scylla serrata</i>	Crab	72	LC50	4629	Rao et al., 1988
Benzene	<i>Scylla serrata</i>	Crab	72	LC50	7339	Rao et al., 1988
Benzene	<i>Scylla serrata</i>	Crab	72	LC50	8941	Rao et al., 1988
Benzene	<i>Scylla serrata</i>	Crab	72	LC50	4629	Rao et al., 1988
Benzene	<i>Scylla serrata</i>	Crab	72	LC50	3291	Rao et al., 1988
Benzene	<i>Scylla serrata</i>	Crab	96	LC50	3678	Rao et al., 1988
Benzene	<i>Scylla serrata</i>	Crab	96	LC50	6090	Rao et al., 1988
Benzene	<i>Scylla serrata</i>	Crab	96	LC50	7682	Rao et al., 1988
Benzene	<i>Therapon jarbua</i>	Tigerfish, crescent perch	24	LC50	96000	Dange and Masurekar, 1984
Benzene	<i>Therapon jarbua</i>	Tigerfish, crescent perch	48	LC50	94000	Dange and Masurekar, 1984
Benzene	<i>Therapon jarbua</i>	Tigerfish, crescent perch	72	LC50	88000	Dange and Masurekar, 1984
Benzene	<i>Therapon jarbua</i>	Tigerfish, crescent perch	96	LC50	84000	Dange and Masurekar, 1984
Benzene	<i>Morone saxatilis</i>	Striped bass	96	LC50	5084	Benville and Korn, 1977
Benzene	<i>Crangon franciscorum</i>	Bay shrimp	96	LC50	17500	Benville and Korn, 1977

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
Benzene	<i>Cancer magister</i>	Dungeness crab, edible	96	LC50	108000	Caldwell et al., 1977
Benzene	<i>Daphnia pulex</i>	Water flea	48	LC50	265000	Canton and Adema, 1978
Benzene	<i>Daphnia pulex</i>	Water flea	48	LC50	345000	Canton and Adema, 1978
Benzene	<i>Daphnia cucullata</i>	Water flea	48	LC50	356000	Canton and Adema, 1978
Benzene	<i>Daphnia cucullata</i>	Water flea	48	LC50	390000	Canton and Adema, 1978
Benzene	<i>Daphnia magna</i>	Water flea	48	LC50	400000	Canton and Adema, 1978
Benzene	<i>Daphnia magna</i>	Water flea	48	LC50	412000	Canton and Adema, 1978
Benzene	<i>Daphnia magna</i>	Water flea	48	LC50	620000	Canton and Adema, 1978
Benzene	<i>Ceriodaphnia dubia</i>	Water flea	168	LC50	12419	Niederlehner et al., 1998
Benzene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	5300	Degraeve et al., 1982
Benzene	<i>Daphnia magna</i>	Water flea	48	LC50	682000	Eastmond et al., 1984
Benzene	<i>Gammarus fossarum</i>	Scud	120	LC50	58000	Erben and Pisl, 1993
Benzene	<i>Gammarus fossarum</i>	Scud	96	LC50	6000	Erben and Pisl, 1993
Benzene	<i>Asellus aquaticus</i>	Aquatic sowbug	96	LC50	290000	Erben and Pisl, 1993
Benzene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	9200	Johnson and Finley, 1980
Benzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	100000	Johnson and Finley, 1980
Benzene	<i>Ictalurus punctatus</i>	Channel catfish	96	LC50	425000	Johnson and Finley, 1980
Benzene	<i>Morone saxatilis</i>	Striped bass	96	LC50	9554	Meyeroff, 1975
Benzene	<i>Oncorhynchus gorbuscha</i>	Pink salmon	96	LC50	4628	Moles et al., 1979
Benzene	<i>Oncorhynchus nerka</i>	Sockeye salmon	96	LC50	4865	Moles et al., 1979
Benzene	<i>Salvelinus malma</i>	Dolly varden	96	LC50	5522	Moles et al., 1979
Benzene	<i>Oncorhynchus gorbuscha</i>	Pink salmon	96	LC50	7424	Moles et al., 1979
Benzene	<i>Oncorhynchus kisutch</i>	Coho salmon,silver salmon	96	LC50	8590	Moles et al., 1979
Benzene	<i>Oncorhynchus nerka</i>	Sockeye salmon	96	LC50	9431	Moles et al., 1979
Benzene	<i>Oncorhynchus tshawytscha</i>	Chinook salmon	96	LC50	10281	Moles et al., 1979
Benzene	<i>Salvelinus malma</i>	Dolly varden	96	LC50	10430	Moles et al., 1979
Benzene	<i>Salvelinus malma</i>	Dolly varden	96	LC50	10482	Moles et al., 1979
Benzene	<i>Oncorhynchus kisutch</i>	Coho salmon,silver salmon	96	LC50	12350	Moles et al., 1979
Benzene	<i>Thymallus arcticus</i>	Arctic grayling	96	LC50	12894	Moles et al., 1979
Benzene	<i>Cottus cognatus</i>	Slimy sculpin	96	LC50	13507	Moles et al., 1979
Benzene	<i>Oncorhynchus gorbuscha</i>	Pink salmon	96	LC50	14980	Moles et al., 1979
Benzene	<i>Gasterosteus aculeatus</i>	Threespine stickleback	96	LC50	21763	Moles et al., 1979

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
Benzene	<i>Oncorhynchus gorbuscha</i>	Pink salmon	96	LC50	297000	Moles et al., 1979
Benzene	<i>Oncorhynchus kisutch</i>	Coho salmon, silver salmon	96	LC50	475000	Moles et al., 1979
Benzene	<i>Solea solea</i>	Dover sole	96	LC50	10070	Pickering and Henderson, 1966
Benzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	22490	Pickering and Henderson, 1966
Benzene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	32000	Pickering and Henderson, 1966
Benzene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	33470	Pickering and Henderson, 1966
Benzene	<i>Carrasius auratus</i>	Goldfish	96	LC50	34420	Pickering and Henderson, 1966
Benzene	<i>Poecilia reticulata</i>	Guppy	96	LC50	36600	Pickering and Henderson, 1966
Benzene	<i>Clupea harengus pallasii</i>	Pacific herring	96	LC50	40000	Struhsaker et al., 1974
Benzene	<i>Palaemonetes pugio</i>	Grass shrimp	96	LC50	27000	Tatem et al., 1978
Benzene	<i>Daphnia pulex</i>	Water flea	96	LC50	15000	Trucco et al., 1983
Benzene	<i>Gambusia affinis</i>	Western mosquitofish	96	LC50	386000	Wallen et al., 1957
Benzene	<i>Pimephales promelas</i>	Fathead minnow	168	LOEC	17200	Marchini et al., 1992
Benzene	<i>Pimephales promelas</i>	Fathead minnow	168	NOEC	10200	Marchini et al., 1992
Benzo(a)pyrene	<i>Daphnia pulex</i>	Water flea	96	LC50	5	Trucco et al., 1983
Benzo(a)pyrene	<i>Pimephales promelas</i>	Fathead minnow	40	LT50	6	Oris, 1987
Chrysene	<i>Daphnia magna</i>	Water flea	24	LT50	0.7	Newsted and Giesy, 1987
Dibenzothiophene	<i>Palaemonetes pugio</i>	Grass shrimp	48	LC50	280	Wofford and Neff, 1978
Ethynylbenzene	<i>Ceriodaphnia dubia</i>	Water flea	168	LOEC	0.13	Tatarazako et al., 2002
Ethynylbenzene	<i>Ceriodaphnia dubia</i>	Water flea	168	NOEC	0.06	Tatarazako et al., 2002
Ethylbenzene	<i>Americamysis bahia</i>	Opossum shrimp	72	LC50	4000	Masten et al., 1994
Ethylbenzene	<i>Americamysis bahia</i>	Opossum shrimp	96	LC50	2600	Masten et al., 1994
Ethylbenzene	<i>Menidia menidia</i>	Atlantic silverside	24	LC50	7000	Masten et al., 1994
Ethylbenzene	<i>Menidia menidia</i>	Atlantic silverside	48	LC50	6400	Masten et al., 1994
Ethylbenzene	<i>Menidia menidia</i>	Atlantic silverside	72	LC50	5800	Masten et al., 1994
Ethylbenzene	<i>Menidia menidia</i>	Atlantic silverside	96	LC50	5100	Masten et al., 1994
Ethylbenzene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	9090	Geiger et al., 1990
Ethylbenzene	<i>Artemia salina</i>	Brine shrimp	24	LC50	11326	Abernethy et al., 1986
Ethylbenzene	<i>Artemia sp.</i>	Brine shrimp	48	LC50	8780	MacLean and Doe, 1989
Ethylbenzene	<i>Artemia sp.</i>	Brine shrimp	48	LC50	13300	MacLean and Doe, 1989
Ethylbenzene	<i>Daphnia magna</i>	Water flea	24	LC50	77000	LeBlanc, 1980
Ethylbenzene	<i>Daphnia magna</i>	Water flea	48	LC50	75000	LeBlanc, 1980

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
Ethylbenzene	<i>Daphnia magna</i>	Water flea	48	LC50	18400	MacLean and Doe, 1989
Ethylbenzene	<i>Daphnia magna</i>	Water flea	48	LC50	13900	MacLean and Doe, 1989
Ethylbenzene	<i>Gammarus pseudolimnaeus</i>	Scud	96	LC50	1940	Brooke, 1987
Ethylbenzene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	169000	Buccafusco et al., 1981
Ethylbenzene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	90000	Mayer and Ellersieck, 1986
Ethylbenzene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	100000	Mayer and Ellersieck, 1986
Ethylbenzene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	160000	Mayer and Ellersieck, 1986
Ethylbenzene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	135000	Mayer and Ellersieck, 1986
Ethylbenzene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	134000	Mayer and Ellersieck, 1986
Ethylbenzene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	80000	Mayer and Ellersieck, 1986
Ethylbenzene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	135000	Mayer and Ellersieck, 1986
Ethylbenzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	150000	Buccafusco et al., 1981
Ethylbenzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	88000	Mayer and Ellersieck, 1986
Ethylbenzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	84000	Mayer and Ellersieck, 1986
Ethylbenzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	140000	Mayer and Ellersieck, 1986
Ethylbenzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	56000	Mayer and Ellersieck, 1986
Ethylbenzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	86000	Mayer and Ellersieck, 1986
Ethylbenzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	135000	Mayer and Ellersieck, 1986
Ethylbenzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	134000	Mayer and Ellersieck, 1986
Ethylbenzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	80000	Mayer and Ellersieck, 1986
Ethylbenzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	135000	Mayer and Ellersieck, 1986
Ethylbenzene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	24	LC50	14000	Mayer and Ellersieck, 1986
Ethylbenzene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	4200	Galassi et al., 1988
Ethylbenzene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	14000	Mayer and Ellersieck, 1986
Ethylbenzene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	12100	Geiger et al., 1986
Ethylbenzene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	11900	Brooke, 1987
Ethylbenzene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	9100	Brooke, 1987
Ethylbenzene	<i>Poecilia reticulata</i>	Guppy	96	LC50	9600	Galassi et al., 1988
Ethylbenzene	<i>Morone saxatilis</i>	Striped bass	96	LC50	3769	Benville and Korn, 1977
Ethylbenzene	<i>Crangon franciscorum</i>	Bay shrimp	96	LC50	429	Benville and Korn, 1977
Ethylbenzene	<i>Cancer magister</i>	Dungeness crab, edible	96	LC50	13000	Caldwell et al., 1977
Ethylbenzene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	14000	Johnson and Finley, 1980

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
Ethylbenzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	88000	Johnson and Finley, 1980
Ethylbenzene	<i>Ceriodaphnia dubia</i>	Water flea	168	LC50	3611	Neiderlehner et al., 1998
Ethylbenzene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	32000	Pickering and Henderson, 1966
Ethylbenzene	<i>Pimpeles promelas</i>	Fathead minnow	96	LC50	42330	Pickering and Henderson, 1966
Ethylbenzene	<i>Pimpeles promelas</i>	Fathead minnow	96	LC50	48510	Pickering and Henderson, 1966
Ethylbenzene	<i>Carassius auratus</i>	Goldfish	96	LC50	94440	Pickering and Henderson, 1966
Ethylbenzene	<i>Poecilia reticulata</i>	Guppy	96	LC50	97100	Pickering and Henderson, 1966
Ethylbenzene	<i>Mysidopsis bahia</i>	Opossum shrimp	96	LC50	87600	USEPA, 1978
Fluoranthene	<i>Americamysis bahia</i>	Opossum shrimp	48	LC50	5.32	Pelletier et al., 1997
Fluoranthene	<i>Americamysis bahia</i>	Opossum shrimp	48	LC50	63.8	Pelletier et al., 1997
Fluoranthene	<i>Americamysis bahia</i>	Opossum shrimp	96	LC50	40	USEPA, 1978
Fluoranthene	<i>Americamysis bahia</i>	Opossum shrimp	96	LC50	22	Spehar et al., 1999
Fluoranthene	<i>Americamysis bahia</i>	Opossum shrimp	96	LC50	1.4	Spehar et al., 1999
Fluoranthene	<i>Americamysis bahia</i>	Opossum shrimp	96	LC50	31	Spehar et al., 1999
Fluoranthene	<i>Americamysis bahia</i>	Opossum shrimp	96	LC50	58	Spehar et al., 1999
Fluoranthene	<i>Americamysis bahia</i>	Opossum shrimp	96	LC50	12	Spehar et al., 1999
Fluoranthene	<i>Americamysis bahia</i>	Opossum shrimp	96	LC50	12	Spehar et al., 1999
Fluoranthene	<i>Americamysis bahia</i>	Opossum shrimp	96	LC50	2.8	Spehar et al., 1999
Fluoranthene	<i>Americamysis bahia</i>	Opossum shrimp	96	LC50	0.8	Spehar et al., 1999
Fluoranthene	<i>Ampelisca abdita</i>	Amphipod	96	LC50	67	Spehar et al., 1999
Fluoranthene	<i>Arbacia punctulata</i>	Purple-spined sea urchin	48	LC50	3.9	Spehar et al., 1999
Fluoranthene	<i>Arbacia punctulata</i>	Purple-spined sea urchin	96	LC50	1	Spehar et al., 1999
Fluoranthene	<i>Ceriodaphnia dubia</i>	Water flea	48	LC50	45	Oris et al., 1991
Fluoranthene	<i>Chironomus riparius</i>	Midge	264	LC50	64.1	Stewart and Thompson, 1995
Fluoranthene	<i>Chironomus riparius</i>	Midge	264	LC50	70.5	Stewart and Thompson, 1995
Fluoranthene	<i>Chironomus riparius</i>	Midge	264	LC50	61.5	Stewart and Thompson, 1995
Fluoranthene	<i>Chironomus riparius</i>	Midge	264	LC50	86.1	Stewart and Thompson, 1995
Fluoranthene	<i>Chironomus tentans</i>	Midge	240	LC50	37.8	Suedel, 1996
Fluoranthene	<i>Chironomus tentans</i>	Midge	240	LC50	23.6	Suedel, 1996
Fluoranthene	<i>Chironomus tentans</i>	Midge	240	LC50	12.6	Hatch, 1999
Fluoranthene	<i>Chironomus thummi</i>	Midge	48	LC50	44	Horne and Oblad, 1983
Fluoranthene	<i>Corophium insidiosum</i>	Scud	96	LC50	85	Boese et al., 1997

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
Fluoranthene	<i>Corophium spinicorne</i>	Corophiid amphipod	240	LC50	23.9	Swartz et al., 1990
Fluoranthene	<i>Corophium spinicorne</i>	Corophiid amphipod	240	LC50	37.9	Swartz et al., 1990
Fluoranthene	<i>Cyprinodon variegatus</i>	Sheepshead minnow	96	LC50	159	Spehar et al., 1999
Fluoranthene	<i>Cyprinodon variegatus</i>	Sheepshead minnow	96	LC50	0.9	Spehar et al., 1999
Fluoranthene	<i>Daphnia magna</i>	Water flea	240	LC50	102.6	Suedel, 1996
Fluoranthene	<i>Daphnia magna</i>	Water flea	240	LC50	110.5	Suedel, 1996
Fluoranthene	<i>Daphnia magna</i>	Water flea	72	LC50	75.22	Barata and Baird, 2000
Fluoranthene	<i>Daphnia magna</i>	Water flea	72	LC50	73.7	Barata and Baird, 2000
Fluoranthene	<i>Daphnia magna</i>	Water flea	48	LC50	105.7	Suedel, 1996
Fluoranthene	<i>Daphnia magna</i>	Water flea	48	LC50	117	Spehar et al., 1999
Fluoranthene	<i>Daphnia magna</i>	Water flea	48	LC50	1.6	Spehar et al., 1999
Fluoranthene	<i>Daphnia magna</i>	Water flea	48	LC50	78	Spehar et al., 1999
Fluoranthene	<i>Emerita analoga</i>	Pacific sand crab	96	LC50	74	Boese et al., 1997
Fluoranthene	<i>Gammarus minus</i>	Scud	96	LC50	32	Horne and Oblad, 1983
Fluoranthene	<i>Gammarus pseudolimnaeus</i>	Scud	96	LC50	108	Spehar et al., 1999
Fluoranthene	<i>Grandidierella japonica</i>	Scud	96	LC50	36	Boese et al., 1997
Fluoranthene	<i>Homarus americanus</i>	American lobster	96	LC50	24	Spehar et al., 1999
Fluoranthene	<i>Homarus americanus</i>	American lobster	96	LC50	13	Spehar et al., 1999
Fluoranthene	<i>Homarus americanus</i>	American lobster	96	LC50	0.6	Spehar et al., 1999
Fluoranthene	<i>Hyalella azteca</i>	Scud	240	LC50	30.3	Suedel, 1996
Fluoranthene	<i>Hyalella azteca</i>	Scud	240	LC50	60.6	Suedel, 1996
Fluoranthene	<i>Hyalella azteca</i>	Scud	240	LC50	7.3	Hatch, 1999
Fluoranthene	<i>Hyalella azteca</i>	Scud	48	LC50	92.2	Suedel, 1996
Fluoranthene	<i>Hyalella azteca</i>	Scud	96	LC50	44	Spehar et al., 1999
Fluoranthene	<i>Hydra americana</i>	Hydra	96	LC50	70	Spehar et al., 1999
Fluoranthene	<i>Hydra americana</i>	Hydra	96	LC50	2.2	Spehar et al., 1999
Fluoranthene	<i>Hydra americana</i>	Hydra	96	LC50	32	Spehar et al., 1999
Fluoranthene	<i>Ictalurus punctatus</i>	Channel catfish	96	LC50	36	Gendusa, 1990
Fluoranthene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	12.3	Spehar et al., 1999
Fluoranthene	<i>Lumbriculus variegatus</i>	Oligochaete, worm	96	LC50	1.2	Spehar et al., 1999
Fluoranthene	<i>Menidia beryllina</i>	Inland silverside	96	LC50	21	Spehar et al., 1999
Fluoranthene	<i>Menidia beryllina</i>	Inland silverside	96	LC50	30	Spehar et al., 1999

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
Fluoranthene	<i>Menidia beryllina</i>	Inland silverside	96	LC50	103	Spehar et al., 1999
Fluoranthene	<i>Menidia beryllina</i>	Inland silverside	96	LC50	49	Spehar et al., 1999
Fluoranthene	<i>Menidia beryllina</i>	Inland silverside	96	LC50	30	Spehar et al., 1999
Fluoranthene	<i>Menidia beryllina</i>	Inland silverside	96	LC50	2.3	Spehar et al., 1999
Fluoranthene	<i>Mulinia lateralis</i>	Clam	48	LC50	2.8	Spehar et al., 1999
Fluoranthene	<i>Mulinia lateralis</i>	Clam	96	LC50	1.8	Pelletier et al., 1997
Fluoranthene	<i>Nereis arenaceodentata</i>	Polychaete worm	96	LC50	500	Rossi and Neff, 1978
Fluoranthene	<i>Nereis arenaceodentata</i>	Polychaete worm	96	LC50	500	Neff et al., 1976
Fluoranthene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	187	Horne and Oblad, 1983
Fluoranthene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	7.7	Spehar et al., 1999
Fluoranthene	<i>Palaemonetes sp.</i>	Grass shrimp,freshwater prawn	96	LC50	6.5	Spehar et al., 1999
Fluoranthene	<i>Palaemonetes sp.</i>	Grass shrimp,freshwater prawn	96	LC50	142	Spehar et al., 1999
Fluoranthene	<i>Palaemonetes sp.</i>	Grass shrimp,freshwater prawn	96	LC50	22	Spehar et al., 1999
Fluoranthene	<i>Palaemonetes sp.</i>	Grass shrimp,freshwater prawn	96	LC50	3.3	Spehar et al., 1999
Fluoranthene	<i>Physa heterostrophia</i>	Pond snail, pneumonate snail	96	LC50	137	Horne and Oblad, 1983
Fluoranthene	<i>Physella virgata</i>	Snail	96	LC50	82	Spehar et al., 1999
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	720	LC50	7.1	Gendusa, 1990
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	144	LC50	6.83	Diamond et al., 1995
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	95	Horne and Oblad, 1983
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	6.83	Diamond, 1995
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	12.2	Spehar et al., 1999
Fluoranthene	<i>Pleuronectes americanus</i>	Winter flounder	96	LC50	0.1	Spehar et al., 1999
Fluoranthene	<i>Rhepoxynius abronius</i>	Amphipod	240	LC50	22.7	Swartz et al., 1990
Fluoranthene	<i>Rhepoxynius abronius</i>	Amphipod	240	LC50	29.4	Swartz et al., 1990
Fluoranthene	<i>Rhepoxynius abronius</i>	Amphipod	240	LC50	24.2	Swartz et al., 1990
Fluoranthene	<i>Rhepoxynius abronius</i>	Amphipod	240	LC50	11.1	Swartz et al., 1990
Fluoranthene	<i>Americamysis bahia</i>	Opossum shrimp	744	LOEC	18.8	Spehar et al., 1999
Fluoranthene	<i>Daphnia magna</i>	Water flea	504	LOEC	35.3	Spehar et al., 1999
Fluoranthene	<i>Daphnia magna</i>	Water flea	504	LOEC	35.3	Brooke, 1993
Fluoranthene	<i>Daphnia magna</i>	Water flea	504	LOEC	1.5	Spehar et al., 1999
Fluoranthene	<i>Daphnia magna</i>	Water flea	504	LOEC	73.2	Brooke, 1993
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	768	LOEC	21.7	Spehar et al., 1999

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	768	LOEC	21.7	Spehar et al., 1999
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	113	LT50	4.8	Weinstein and Oris, 1999
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	45	LT50	69.75	Diamond, 1995
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	46	LT50	69.75	Diamond, 1995
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	49	LT50	69.75	Diamond, 1995
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	52	LT50	69.75	Diamond, 1995
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	55	LT50	69.75	Diamond, 1995
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	56	LT50	27.6	Diamond, 1995
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	61	LT50	69.75	Diamond, 1995
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	64	LT50	69.75	Diamond, 1995
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	64	LT50	28.2	Diamond, 1995
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	65	LT50	69.75	Diamond, 1995
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	67	LT50	27.85	Diamond, 1995
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	73	LT50	69.75	Diamond, 1995
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	73	LT50	69.75	Diamond, 1995
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	76	LT50	69.75	Diamond, 1995
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	91	LT50	28.2	Diamond, 1995
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	95	LT50	9.7	Weinstein and Oris, 1999
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	97	LT50	28.2	Diamond, 1995
Fluoranthene	<i>Americamysis bahia</i>	Opossum shrimp	744	NOEC	11.1	Spehar et al., 1999
Fluoranthene	<i>Daphnia magna</i>	Water flea	504	NOEC	17	Spehar et al., 1999
Fluoranthene	<i>Daphnia magna</i>	Water flea	504	NOEC	17	Brooke, 1993
Fluoranthene	<i>Daphnia magna</i>	Water flea	72	NOEC	20	Barata and Baird, 2000
Fluoranthene	<i>Daphnia magna</i>	Water flea	504	NOEC	1.4	Spehar et al., 1999
Fluoranthene	<i>Daphnia magna</i>	Water flea	504	NOEC	35.3	Brooke, 1993
Fluoranthene	<i>Daphnia magna</i>	Water flea	72	NOEC	30	Barata and Baird, 2000
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	768	NOEC	10.4	Spehar et al., 1999
Fluoranthene	<i>Pimephales promelas</i>	Fathead minnow	768	NOEC	10.4	Spehar et al., 1999
Isopropylbenzene	<i>Daphnia magna</i>	Water flea	48	LC50	601	Abernethy et al., 1986
Naphthalene	<i>Artemia salina</i>	Brine shrimp	24	LC50	10638	Abernethy et al., 1986
Naphthalene	<i>Artemia salina</i>	Brine shrimp	24	LC50	10638	Abernethy et al., 1986
Naphthalene	<i>Artemia sp.</i>	Brine shrimp	48	LC50	12500	MacLean and Doe, 1989

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
Naphthalene	<i>Artemia sp.</i>	Brine shrimp	48	LC50	9820	MacLean and Doe, 1989
Naphthalene	<i>Callinectes sapidus</i>	Blue crab	24	LC50	1980	Sabourin, 1982
Naphthalene	<i>Callinectes sapidus</i>	Blue crab	24	LC50	2250	Sabourin, 1982
Naphthalene	<i>Callinectes sapidus</i>	Blue crab	24	LC50	3120	Sabourin, 1982
Naphthalene	<i>Chironomus attenuatus</i>	Midge	24	LC50	13100	Darville and Wilhm, 1984
Naphthalene	<i>Chironomus attenuatus</i>	Midge	24	LC50	13000	Darville and Wilhm, 1984
Naphthalene	<i>Chironomus attenuatus</i>	Midge	24	LC50	13900	Darville, 1982
Naphthalene	<i>Chironomus attenuatus</i>	Midge	24	LC50	13300	Darville, 1982
Naphthalene	<i>Chironomus tentans</i>	Midge	48	LC50	2810	Millemann et al., 1984
Naphthalene	<i>Cyprinodon variegatus</i>	Sheepshead minnow	24	LC50	2400	Anderson et al., 1974
Naphthalene	<i>Daphnia magna</i>	Water flea	24	LC50	17000	LeBlanc, 1980
Naphthalene	<i>Daphnia magna</i>	Water flea	48	LC50	8600	LeBlanc, 1980
Naphthalene	<i>Daphnia magna</i>	Water flea	48	LC50	2160	Millemann et al., 1984
Naphthalene	<i>Daphnia magna</i>	Water flea	48	LC50	22600	Eastmond et al., 1984
Naphthalene	<i>Daphnia magna</i>	Water flea	48	LC50	12300	MacLean and Doe, 1989
Naphthalene	<i>Daphnia magna</i>	Water flea	48	LC50	11400	MacLean and Doe, 1989
Naphthalene	<i>Daphnia magna</i>	Water flea	24	LC50	13200	Crider et al., 1982
Naphthalene	<i>Daphnia magna</i>	Water flea	24	LC50	6600	Crider et al., 1982
Naphthalene	<i>Daphnia magna</i>	Water flea	48	LC50	3400	Crider et al., 1982
Naphthalene	<i>Daphnia magna</i>	Water flea	48	LC50	4100	Crider et al., 1982
Naphthalene	<i>Daphnia pulex</i>	Water flea	48	LC50	3405	Geiger, 1982
Naphthalene	<i>Daphnia pulex</i>	Water flea	96	LC50	1000	Trucco et al., 1983
Naphthalene	<i>Elasmopus pectinicus</i>	Scud	24	LC50	3650	Lee and Nicol, 1978
Naphthalene	<i>Elasmopus pectinicus</i>	Scud	48	LC50	2800	Lee and Nicol, 1978
Naphthalene	<i>Elasmopus pectinicus</i>	Scud	96	LC50	2680	Lee and Nicol, 1978
Naphthalene	<i>Eualus suckleyi</i>	Shortscale eualid	96	LC50	1390	Rice and Thomas, 1989
Naphthalene	<i>Eurytemora affinis</i>	Calanoid copepod	24	LC50	3798	Ott et al., 1978
Naphthalene	<i>Gammarus minus</i>	Scud	48	LC50	3930	Millemann et al., 1984
Naphthalene	<i>Hemigrapsus nudus</i>	Shore crab	192	LC50	1950	Gharrett and Rice, 1987
Naphthalene	<i>Macrobrachium kistnensis</i>	Shrimp	24	LC50	5000	Jaiswal et al., 1989
Naphthalene	<i>Macrobrachium kistnensis</i>	Shrimp	24	LC50	5000	Jaiswal et al., 1989
Naphthalene	<i>Macrobrachium kistnensis</i>	Shrimp	24	LC50	5000	Jaiswal et al., 1989

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
Naphthalene	<i>Macrobrachium kistnensis</i>	Shrimp	24	LC50	5000	Jaiswal et al., 1989
Naphthalene	<i>Macrobrachium kistnensis</i>	Shrimp	24	LC50	5000	Jaiswal et al., 1989
Naphthalene	<i>Macrobrachium kistnensis</i>	Shrimp	24	LC50	5000	Jaiswal et al., 1989
Naphthalene	<i>Macrobrachium kistnensis</i>	Shrimp	48	LC50	5000	Jaiswal et al., 1989
Naphthalene	<i>Macrobrachium kistnensis</i>	Shrimp	48	LC50	5000	Jaiswal et al., 1989
Naphthalene	<i>Macrobrachium kistnensis</i>	Shrimp	48	LC50	3000	Jaiswal et al., 1989
Naphthalene	<i>Macrobrachium kistnensis</i>	Shrimp	48	LC50	3000	Jaiswal et al., 1989
Naphthalene	<i>Macrobrachium kistnensis</i>	Shrimp	48	LC50	3000	Jaiswal et al., 1989
Naphthalene	<i>Macrobrachium kistnensis</i>	Shrimp	72	LC50	3000	Jaiswal et al., 1989
Naphthalene	<i>Macrobrachium kistnensis</i>	Shrimp	72	LC50	3000	Jaiswal et al., 1989
Naphthalene	<i>Macrobrachium kistnensis</i>	Shrimp	72	LC50	3000	Jaiswal et al., 1989
Naphthalene	<i>Macrobrachium kistnensis</i>	Shrimp	72	LC50	3000	Jaiswal et al., 1989
Naphthalene	<i>Macrobrachium kistnensis</i>	Shrimp	72	LC50	3000	Jaiswal et al., 1989
Naphthalene	<i>Macrobrachium kistnensis</i>	Shrimp	72	LC50	3000	Jaiswal et al., 1989
Naphthalene	<i>Macrobrachium kistnensis</i>	Shrimp	96	LC50	3000	Jaiswal et al., 1989
Naphthalene	<i>Macrobrachium kistnensis</i>	Shrimp	96	LC50	3000	Jaiswal et al., 1989
Naphthalene	<i>Macrobrachium kistnensis</i>	Shrimp	96	LC50	3000	Jaiswal et al., 1989
Naphthalene	<i>Micropterus salmoides</i>	Largemouth bass	168	LC50	510	Black et al., 1983
Naphthalene	<i>Micropterus salmoides</i>	Largemouth bass	168	LC50	680	Millemann et al., 1984
Naphthalene	<i>Neomysis americana</i>	Opossum shrimp	96	LC50	1280	Smith and Hargreaves, 1983
Naphthalene	<i>Neomysis americana</i>	Opossum shrimp	96	LC50	850	Smith and Hargreaves, 1983
Naphthalene	<i>Nereis arenaceodentata</i>	Polychaete worm	96	LC50	3800	Rossi and Neff, 1978
Naphthalene	<i>Nereis arenaceodentata</i>	Polychaete worm	96	LC50	3800	Neff et al., 1976
Naphthalene	<i>Oncorhynchus gorbuscha</i>	Pink salmon	24	LC50	920	Thomas and Rice, 1978
Naphthalene	<i>Oncorhynchus gorbuscha</i>	Pink salmon	960	LC50	1200	Moles and Rice, 1983
Naphthalene	<i>Oncorhynchus gorbuscha</i>	Pink salmon	48	LC50	960	Rice and Thomas, 1989
Naphthalene	<i>Oncorhynchus gorbuscha</i>	Pink salmon	48	LC50	900	Rice and Thomas, 1989
Naphthalene	<i>Oncorhynchus gorbuscha</i>	Pink salmon	48	LC50	990	Rice and Thomas, 1989
Naphthalene	<i>Oncorhynchus gorbuscha</i>	Pink salmon	48	LC50	1010	Rice and Thomas, 1989
Naphthalene	<i>Oncorhynchus gorbuscha</i>	Pink salmon	48	LC50	890	Rice and Thomas, 1989
Naphthalene	<i>Oncorhynchus gorbuscha</i>	Pink salmon	96	LC50	1370	Korn et al., 1979
Naphthalene	<i>Oncorhynchus gorbuscha</i>	Pink salmon	96	LC50	1840	Korn et al., 1979

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
Naphthalene	<i>Oncorhynchus gorbuscha</i>	Pink salmon	96	LC50	1240	Korn et al., 1979
Naphthalene	<i>Oncorhynchus gorbuscha</i>	Pink salmon	96	LC50	1200	Moles and Rice, 1983
Naphthalene	<i>Oncorhynchus gorbuscha</i>	Pink salmon	24	LC50	1560	Korn et al., 1979
Naphthalene	<i>Oncorhynchus gorbuscha</i>	Pink salmon	24	LC50	1840	Korn et al., 1979
Naphthalene	<i>Oncorhynchus gorbuscha</i>	Pink salmon	24	LC50	1380	Korn et al., 1979
Naphthalene	<i>Oncorhynchus kisutch</i>	Coho salmon,silver salmon	96	LC50	3220	Moles, 1980
Naphthalene	<i>Oncorhynchus kisutch</i>	Coho salmon,silver salmon	96	LC50	2100	Moles et al., 1981
Naphthalene	<i>Oncorhynchus kisutch</i>	Coho salmon,silver salmon	96	LC50	5600	Korn and Rice, 1981
Naphthalene	<i>Oncorhynchus kisutch</i>	Coho salmon,silver salmon	96	LC50	2100	USEPA, 2000
Naphthalene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	552	LC50	120	Black et al., 1983
Naphthalene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	648	LC50	110	Black et al., 1983
Naphthalene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	1800	Edsall, 1991
Naphthalene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	6100	Edsall, 1991
Naphthalene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	2600	Edsall, 1991
Naphthalene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	4400	Edsall, 1991
Naphthalene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	5500	Edsall, 1991
Naphthalene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	4500	Edsall, 1991
Naphthalene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	1600	Degraeve et al., 1982
Naphthalene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	2250	Bergman and Anderson, 1977
Naphthalene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	648	LC50	120	Millemann et al., 1984
Naphthalene	<i>Palaemonetes pugio</i>	Daggerblade grass shrimp	24	LC50	2600	Anderson et al., 1974
Naphthalene	<i>Palaemonetes pugio</i>	Daggerblade grass shrimp	24	LC50	2600	Tatem, 1975
Naphthalene	<i>Palaemonetes pugio</i>	Daggerblade grass shrimp	48	LC50	2600	Tatem, 1975
Naphthalene	<i>Palaemonetes pugio</i>	Daggerblade grass shrimp	96	LC50	2350	Tatem et al., 1978
Naphthalene	<i>Palaemonetes pugio</i>	Daggerblade grass shrimp	96	LC50	2350	Tatem, 1975
Naphthalene	<i>Palaemonetes pugio</i>	Daggerblade grass shrimp	96	LC50	2400	Neff et al., 1976
Naphthalene	<i>Palaemonetes pugio</i>	Daggerblade grass shrimp	48	LC50	2350	Tatem and Anderson, 1973
Naphthalene	<i>Pandalus goniurus</i>	Humpy shrimp	96	LC50	2160	Korn et al., 1979
Naphthalene	<i>Pandalus goniurus</i>	Humpy shrimp	96	LC50	1020	Korn et al., 1979
Naphthalene	<i>Pandalus goniurus</i>	Humpy shrimp	96	LC50	971	Korn et al., 1979
Naphthalene	<i>Pandalus goniurus</i>	Humpy shrimp	24	LC50	2210	Korn et al., 1979
Naphthalene	<i>Pandalus goniurus</i>	Humpy shrimp	24	LC50	2060	Korn et al., 1979

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
Naphthalene	<i>Pandalus goniurus</i>	Humpy shrimp	24	LC50	1290	Korn et al., 1979
Naphthalene	<i>Penaeus aztecus</i>	Brown shrimp	24	LC50	2500	Anderson et al., 1974
Naphthalene	<i>Penaeus aztecus</i>	Brown shrimp	96	LC50	2500	Tatem et al., 1978
Naphthalene	<i>Physa gyrina</i>	Pouch snail	48	LC50	5020	Millemann et al., 1984
Naphthalene	<i>Pimephales promelas</i>	Fathead minnow	24	LC50	7760	Holcombe et al., 1984
Naphthalene	<i>Pimephales promelas</i>	Fathead minnow	48	LC50	6350	Holcombe et al., 1984
Naphthalene	<i>Pimephales promelas</i>	Fathead minnow	72	LC50	6080	Holcombe et al., 1984
Naphthalene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	6080	Holcombe et al., 1984
Naphthalene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	1990	Millemann et al., 1984
Naphthalene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	6140	Geiger et al., 1985
Naphthalene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	6140	Broderius et al., 1995
Naphthalene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	7900	Degraeve et al., 1982
Naphthalene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	4900	Bergman and Anderson, 1977
Naphthalene	<i>Tanytarsus dissimilis</i>	Midge	48	LC50	20700	Darville and Wilhm, 1984
Naphthalene	<i>Tanytarsus dissimilis</i>	Midge	48	LC50	12600	Darville and Wilhm, 1984
Naphthalene	<i>Tanytarsus dissimilis</i>	Midge	48	LC50	13700	Darville, 1982
Naphthalene	<i>Tanytarsus dissimilis</i>	Midge	48	LC50	12200	Darville, 1982
Naphthalene	<i>Therapon jarbua</i>	Tigerfish, crescent perch	24	LC50	22500	Dange and Masurekar, 1984
Naphthalene	<i>Therapon jarbua</i>	Tigerfish, crescent perch	48	LC50	20000	Dange and Masurekar, 1984
Naphthalene	<i>Therapon jarbua</i>	Tigerfish, crescent perch	72	LC50	18000	Dange and Masurekar, 1984
Naphthalene	<i>Therapon jarbua</i>	Tigerfish, crescent perch	96	LC50	15500	Dange and Masurekar, 1984
Naphthalene	<i>Tilapia mossambica</i>	Mozambique tilapia	96	LC50	7900	Dange, 1986
Naphthalene	<i>Daphnia pulex</i>	Water flea	48	LC50	2920	Geiger and Buikema, 1981
Naphthalene	<i>Daphnia magna</i>	Water flea	48	LC50	4730	Abernethy et al., 1986
Naphthalene	<i>Neomysis americana</i>	Mysid	96	LC50	850	Smith and Hargreaves, 1984
Naphthalene	<i>Neomysis americana</i>	Mysid	96	LC50	1280	Smith and Hargreaves, 1984
Naphthalene	<i>Daphnia pulex</i>	Water flea	96	LC50	1000	Trucco et al., 1983
Naphthalene	<i>Oncorhynchus kisutch</i>	Coho salmon, silver salmon	96	LC50	2100	USEPA, 1995
Naphthalene	<i>Daphnia magna</i>	Water flea	48	LC50	8570	USEPA, 1978
Naphthalene	<i>Oncorhynchus gorbuscha</i>	Pink salmon	24	LC50	920	Thomas and Rice, 1978
Phenanthrene	<i>Daphnia magna</i>	Water flea	24	EC50	302	Brooke, 1994
Phenanthrene	<i>Daphnia magna</i>	Water flea	48	EC50	212	Brooke, 1994

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
Phenanthrene	<i>Daphnia magna</i>	Water flea	48	EC50	230	Brooke, 1994
Phenanthrene	<i>Artemia salina</i>	Brine shrimp	24	LC50	677	Abernethy et al., 1986
Phenanthrene	<i>Artemia salina</i>	Brine shrimp	24	LC50	677	Abernethy et al., 1986
Phenanthrene	<i>Chironomus tentans</i>	Midge	48	LC50	490	Millemann et al., 1984
Phenanthrene	<i>Cyprinodon variegatus</i>	Sheepshead minnow	96	LC50	478	Moreau et al., 1999
Phenanthrene	<i>Daphnia magna</i>	Water flea	48	LC50	700	Millemann et al., 1984
Phenanthrene	<i>Daphnia magna</i>	Water flea	48	LC50	843	Eastmond et al., 1984
Phenanthrene	<i>Daphnia pulex</i>	Water flea	48	LC50	1120	Geiger, 1982
Phenanthrene	<i>Daphnia pulex</i>	Water flea	96	LC50	100	Trucco et al., 1983
Phenanthrene	<i>Eohaustorius estuarius</i>	Amphipod	240	LC50	158	Swartz et al., 1995
Phenanthrene	<i>Gammarus minus</i>	Scud	48	LC50	460	Millemann et al., 1984
Phenanthrene	<i>Leptocheirus plumulosus</i>	Amphipod	240	LC50	180	Dewitt et al., 1992
Phenanthrene	<i>Micropterus salmoides</i>	Largemouth bass	168	LC50	180	Black et al., 1983
Phenanthrene	<i>Micropterus salmoides</i>	Largemouth bass	168	LC50	250	Millemann et al., 1984
Phenanthrene	<i>Nereis arenaceodentata</i>	Polychaete worm	336	LC50	501	Emery and Dillon, 1996
Phenanthrene	<i>Nereis arenaceodentata</i>	Polychaete worm	336	LC50	501	Emery and Dillon, 1996
Phenanthrene	<i>Nereis arenaceodentata</i>	Polychaete worm	96	LC50	51	Emery and Dillon, 1996
Phenanthrene	<i>Nereis arenaceodentata</i>	Polychaete worm	96	LC50	51	Emery and Dillon, 1996
Phenanthrene	<i>Nereis arenaceodentata</i>	Polychaete worm	96	LC50	600	Rossi and Neff, 1978
Phenanthrene	<i>Nereis arenaceodentata</i>	Polychaete worm	96	LC50	600	Neff et al., 1976
Phenanthrene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	552	LC50	40	Black et al., 1983
Phenanthrene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	648	LC50	40	Black et al., 1983
Phenanthrene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	1440	LC50	0.0002	Passino-Reader, 1993
Phenanthrene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	648	LC50	30	Millemann et al., 1984
Phenanthrene	<i>Daphnia magna</i>	Water flea	504	LOEC	93	Brooke, 1993
Phenanthrene	<i>Daphnia magna</i>	Water flea	504	LOEC	345	Brooke, 1993
Phenanthrene	<i>Daphnia pulex</i>	Water flea	384	LOEC	60	Savino and Tanabe, 1989
Phenanthrene	<i>Daphnia pulex</i>	Water flea	384	LOEC	60	Savino and Tanabe, 1989
Phenanthrene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	1440	LOEC	44	Passino-Reader et al., 1995
Phenanthrene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	1440	LOEC	88	Passino-Reader et al., 1995
Phenanthrene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	1440	LOEC	75	Passino-Reader et al., 1995
Phenanthrene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	1440	LOEC	75	Passino-Reader et al., 1995

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
Phenanthrene	<i>Daphnia magna</i>	Water flea	504	NOEC	48	Brooke, 1993
Phenanthrene	<i>Daphnia magna</i>	Water flea	504	NOEC	183	Brooke, 1993
Phenanthrene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	1440	NOEC	44	Passino-Reader et al., 1995
Phenanthrene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	1440	NOEC	19	Passino-Reader et al., 1995
Phenanthrene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	1440	NOEC	38	Passino-Reader et al., 1995
Phenanthrene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	1440	NOEC	38	Passino-Reader et al., 1995
Propyl benzene	<i>Oncorhynchus mykiss</i>	Rainbow trout	96	LC50	1550	Galassi et al., 1988
Pyrene	<i>Americamysis bahia</i>	Opossum shrimp	48	LC50	0.89	Pelletier et al., 1997
Pyrene	<i>Americamysis bahia</i>	Opossum shrimp	48	LC50	24.8	Pelletier et al., 1997
Pyrene	<i>Mulinia lateralis</i>	Clam	96	LC50	1.68	Pelletier et al., 1997
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	25000	Devlin, 1983
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	55000	Devlin, 1983
Toluene	<i>Morone saxatilis</i>	Striped bass	96	LC50	6398	Benville and Korn, 1977
Toluene	<i>Crangon franciscorum</i>	Bay shrimp	96	LC50	3769	Benville and Korn, 1977
Toluene	<i>Carrisius auratus</i>	Goldfish	96	LC50	22800	Brenniman et al., 1976
Toluene	<i>Oncorhynchus mykiss</i>	Rainbow trout	96	LC50	6780	Brooke et al., 1986
Toluene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	13000	Buccafusco et al., 1981
Toluene	<i>Cancer magister</i>	Dungeness, Edible crab	96	LC50	28000	Caldwell et al., 1977
Toluene	<i>Therapon jarbua</i>	Tigerfish crescent perch	96	LC50	128000	Dange and Masurekar, 1984
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	18000	Devlin et al., 1982
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	25000	Devlin et al., 1982
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	26000	Devlin et al., 1982
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	27000	Devlin et al., 1982
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	28000	Devlin et al., 1982
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	30000	Devlin et al., 1982
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	31000	Devlin et al., 1982
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	36000	Devlin et al., 1982
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	55000	Devlin et al., 1982
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	59000	Devlin et al., 1982
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	66000	Devlin et al., 1982
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	72000	Devlin et al., 1982
Toluene	<i>Oncorhynchus mykiss</i>	Rainbow trout	96	LC50	5800	Galassi et al., 1988

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
Toluene	<i>Poecilia reticulata</i>	Guppy	96	LC50	28200	Galassi et al., 1988
Toluene	<i>Pimpeles promelas</i>	Fathead minnow	96	LC50	36200	Geiger et al., 1986
Toluene	<i>Pimpeles promelas</i>	Fathead minnow	96	LC50	31700	Geiger et al., 1990
Toluene	<i>Clarias lazera</i>	Catfish	96	LC50	26200	Ghazaly, 1991
Toluene	<i>Oncorhynchus mykiss</i>	Rainbow trout	96	LC50	24000	Johnson and Finley, 1980
Toluene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	170000	Johnson and Finley, 1980
Toluene	<i>Ictalurus punctatus</i>	Channel catfish	96	LC50	240000	Johnson and Finley, 1980
Toluene	<i>Oncorhynchus gorbuscha</i>	Pink salmon	96	LC50	6410	Korn et al., 1979
Toluene	<i>Oncorhynchus gorbuscha</i>	Pink salmon	96	LC50	7630	Korn et al., 1979
Toluene	<i>Oncorhynchus gorbuscha</i>	Pink salmon	96	LC50	8090	Korn et al., 1979
Toluene	<i>Oncorhynchus kiustch</i>	Coho salmon	96	LC50	9360	Korn et al., 1979
Toluene	<i>Eualus sp.</i>	Shrimp	96	LC50	14700	Korn et al., 1979
Toluene	<i>Eualus sp.</i>	Shrimp	96	LC50	20200	Korn et al., 1979
Toluene	<i>Eualus sp.</i>	Shrimp	96	LC50	21400	Korn et al., 1979
Toluene	<i>Daphnia magna</i>	Water flea	48	LC50	310000	LeBlanc, 1980
Toluene	<i>Ceriodaphnia dubia</i>	Water flea	168	LC50	3409	Niederlehner et al., 1998
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	9390	Marchini et al., 1992
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	17030	Marchini et al., 1992
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	36200	Marchini et al., 1992
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	54000	Mayes et al., 1983
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	56400	Mayes et al., 1983
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	77400	Mayes et al., 1983
Toluene	<i>Oncorhynchus kiustch</i>	Coho Salmon	96	LC50	5500	Moles et al., 1981
Toluene	<i>Oncorhynchus kiustch</i>	Coho Salmon	96	LC50	8110	Moles, 1980
Toluene	<i>Cyclops viridis</i>	Cyclopoid copepod	96	LC50	215000	Panigrahi and Konar, 1989
Toluene	<i>Pimpeles promelas</i>	Fathead minnow	96	LC50	12600	Pearson et al., 1979
Toluene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	24000	Pickering and Henderson, 1966
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	34270	Pickering and Henderson, 1966
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	42330	Pickering and Henderson, 1966
Toluene	<i>Carrisius auratus</i>	Goldfish	96	LC50	57680	Pickering and Henderson, 1966
Toluene	<i>Poecilia reticulata</i>	Guppy	96	LC50	59300	Pickering and Henderson, 1966
Toluene	<i>Eualus suckleyi</i>	Shrimp	96	LC50	19750	Rice and Thomas, 1989

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
Toluene	<i>Diaptomus forbesi</i>	Calanoid copepod	96	LC50	447000	Saha and Konar, 1983
Toluene	<i>Oryzias latipes</i>	Medaka, high eyes	96	LC50	54000	Stoss and Haines, 1979
Toluene	<i>Palaemonetes pugio</i>	Grass shrimp	96	LC50	9500	Tatem et al., 1978
Toluene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	18000	Devlin, 1983
Xylene	<i>Artemia salina</i>	Brine shrimp	24	LC50	5.7	Calleja et al., 1994
Xylene	<i>Brachionus calyciflorus</i>	Rotifer	24	LC50	9.2	Calleja et al., 1994
Xylene	<i>Carassius auratus</i>	Goldfish	24	LC50	75000	Jensen, 1978
Xylene	<i>Carassius auratus</i>	Goldfish	24	LC50	30550	Brenniman et al., 1976
Xylene	<i>Carassius auratus</i>	Goldfish	24	LC50	36810	Pickering and Henderson, 1966
Xylene	<i>Carassius auratus</i>	Goldfish	48	LC50	25100	Brenniman et al., 1976
Xylene	<i>Carassius auratus</i>	Goldfish	48	LC50	36810	Pickering and Henderson, 1966
Xylene	<i>Carassius auratus</i>	Goldfish	72	LC50	20720	Brenniman et al., 1976
Xylene	<i>Carassius auratus</i>	Goldfish	96	LC50	16940	Brenniman et al., 1976
Xylene	<i>Carassius auratus</i>	Goldfish	96	LC50	36810	Pickering and Henderson, 1966
Xylene	<i>Danio rerio</i>	Zebra danio	48	LC50	20000	Slooff, 1979
Xylene	<i>Danio rerio</i>	Zebra danio	48	LC50	20000	Slooff, 1978
Xylene	<i>Diaptomus forbesi</i>	Calanoid copepod	96	LC50	99500	Saha and Konar, 1983
Xylene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	25600	Bailey et al., 1985
Xylene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	16800	Bailey et al., 1985
Xylene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	10400	Bailey et al., 1985
Xylene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	24000	Pickering and Henderson, 1966
Xylene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	36000	Cope, 1965
Xylene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	14000	Mayer and Ellersieck, 1986
Xylene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	12500	Mayer and Ellersieck, 1986
Xylene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	16500	Mayer and Ellersieck, 1986
Xylene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	14000	Mayer and Ellersieck, 1986
Xylene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	12000	Mayer and Ellersieck, 1986
Xylene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	12000	Mayer and Ellersieck, 1986
Xylene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	16300	Mayer and Ellersieck, 1986
Xylene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	17400	Mayer and Ellersieck, 1986
Xylene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	15000	Mayer and Ellersieck, 1986
Xylene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	15600	Mayer and Ellersieck, 1986

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
Xylene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	13500	Mayer and Ellersieck, 1986
Xylene	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	15600	Mayer and Ellersieck, 1986
Xylene	<i>Lepomis macrochirus</i>	Bluegill	48	LC50	25600	Bailey et al., 1985
Xylene	<i>Lepomis macrochirus</i>	Bluegill	48	LC50	16500	Bailey et al., 1985
Xylene	<i>Lepomis macrochirus</i>	Bluegill	48	LC50	24000	Pickering and Henderson, 1966
Xylene	<i>Lepomis macrochirus</i>	Bluegill	48	LC50	19000	Cope, 1965
Xylene	<i>Lepomis macrochirus</i>	Bluegill	72	LC50	25600	Bailey et al., 1985
Xylene	<i>Lepomis macrochirus</i>	Bluegill	72	LC50	16500	Bailey et al., 1985
Xylene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	24500	Bailey et al., 1985
Xylene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	15700	Bailey et al., 1985
Xylene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	20870	Pickering and Henderson, 1966
Xylene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	19000	Cope, 1965
Xylene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	13500	Mayer and Ellersieck, 1986
Xylene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	8600	Mayer and Ellersieck, 1986
Xylene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	12000	Mayer and Ellersieck, 1986
Xylene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	13300	Mayer and Ellersieck, 1986
Xylene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	12000	Mayer and Ellersieck, 1986
Xylene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	12000	Mayer and Ellersieck, 1986
Xylene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	16100	Mayer and Ellersieck, 1986
Xylene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	17400	Mayer and Ellersieck, 1986
Xylene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	15000	Mayer and Ellersieck, 1986
Xylene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	14400	Mayer and Ellersieck, 1986
Xylene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	13500	Mayer and Ellersieck, 1986
Xylene	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	15000	Mayer and Ellersieck, 1986
Xylene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	24	LC50	8300	Mayer and Ellersieck, 1986
Xylene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	24	LC50	3300	Mayer and Ellersieck, 1986
Xylene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	24	LC50	13500	Mayer and Ellersieck, 1986
Xylene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	24	LC50	17300	Mayer and Ellersieck, 1986
Xylene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	8200	Mayer and Ellersieck, 1986
Xylene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	3300	Mayer and Ellersieck, 1986
Xylene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	13500	Mayer and Ellersieck, 1986
Xylene	<i>Oncorhynchus mykiss</i>	Rainbow trout,donaldson trout	96	LC50	17300	Mayer and Ellersieck, 1986

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
Xylene	<i>Palaemonetes pugio</i>	Daggerblade grass shrimp	24	LC50	14000	Tatem et al., 1978
Xylene	<i>Palaemonetes pugio</i>	Daggerblade grass shrimp	24	LC50	14000	Tatem, 1975
Xylene	<i>Palaemonetes pugio</i>	Daggerblade grass shrimp	48	LC50	8500	Tatem et al., 1978
Xylene	<i>Palaemonetes pugio</i>	Daggerblade grass shrimp	48	LC50	8500	Tatem, 1975
Xylene	<i>Palaemonetes pugio</i>	Daggerblade grass shrimp	96	LC50	7400	Tatem et al., 1978
Xylene	<i>Palaemonetes pugio</i>	Daggerblade grass shrimp	96	LC50	7400	Tatem, 1975
Xylene	<i>Palaemonetes pugio</i>	Daggerblade grass shrimp	96	LC50	7400	Neff et al., 1976
Xylene	<i>Pimephales promelas</i>	Fathead minnow	24	LC50	28770	Pickering and Henderson, 1966
Xylene	<i>Pimephales promelas</i>	Fathead minnow	24	LC50	28770	Pickering and Henderson, 1966
Xylene	<i>Pimephales promelas</i>	Fathead minnow	24	LC50	42000	Mattson et al., 1976
Xylene	<i>Pimephales promelas</i>	Fathead minnow	48	LC50	27710	Pickering and Henderson, 1966
Xylene	<i>Pimephales promelas</i>	Fathead minnow	48	LC50	28770	Pickering and Henderson, 1966
Xylene	<i>Pimephales promelas</i>	Fathead minnow	48	LC50	42000	Mattson et al., 1976
Xylene	<i>Pimephales promelas</i>	Fathead minnow	72	LC50	42000	Mattson et al., 1976
Xylene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	13400	Geiger et al., 1990
Xylene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	26700	Pickering and Henderson, 1966
Xylene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	28770	Pickering and Henderson, 1966
Xylene	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	42000	Mattson et al., 1976
Xylene	<i>Poecilia reticulata</i>	Guppy	24	LC50	34730	Pickering and Henderson, 1966
Xylene	<i>Poecilia reticulata</i>	Guppy	48	LC50	34730	Pickering and Henderson, 1966
Xylene	<i>Poecilia reticulata</i>	Guppy	96	LC50	34730	Pickering and Henderson, 1966
Xylene	<i>Streptocephalus proboscideus</i>	Fairy shrimp	24	LC50	2.6	Calleja et al., 1994
Xylene	<i>Therapon jarbua</i>	Tigerfish, crescent perch	24	LC50	102000	Dange and Masurekar, 1984
Xylene	<i>Therapon jarbua</i>	Tigerfish, crescent perch	48	LC50	95000	Dange and Masurekar, 1984
Xylene	<i>Therapon jarbua</i>	Tigerfish, crescent perch	72	LC50	92000	Dange and Masurekar, 1984
Xylene	<i>Therapon jarbua</i>	Tigerfish, crescent perch	96	LC50	89000	Dange and Masurekar, 1984
Xylene	<i>Brachionus calyciflorus</i>	Rotifer	48	LOEC	40000	Snell and Moffat, 1992
Xylene	<i>Moina macrocopa</i>	Water flea	180	LT50	10	Wong et al., 1995
Xylene	<i>Brachionus calyciflorus</i>	Rotifer	48	NOEC	20000	Snell and Moffat, 1992

Table A-4. Aliphatic Hydrocarbon Toxicity Data (pages A-28 and A-29)

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
1,1-Dimethylcyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	96	LC50	5200	Benville et al., 1985
1,1-Dimethylcyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	96	LC50	5200	Benville et al., 1985
1,1-Dimethylcyclohexane	<i>Morone saxatilis</i>	Striped bass	96	LC50	6900	Benville et al., 1985
1,1-Dimethylcyclohexane	<i>Morone saxatilis</i>	Striped bass	96	LC50	6900	Benville et al., 1985
1,2-Dimethylcyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	24	LC50	1500	Benville et al., 1985
1,2-Dimethylcyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	96	LC50	1000	Benville et al., 1985
1,2-Dimethylcyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	24	LC50	1500	Benville et al., 1985
1,2-Dimethylcyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	96	LC50	1000	Benville et al., 1985
1,2-Dimethylcyclohexane	<i>Morone saxatilis</i>	Striped bass	96	LC50	3200	Benville et al., 1985
1,2-Dimethylcyclohexane	<i>Morone saxatilis</i>	Striped bass	96	LC50	3200	Benville et al., 1985
1,3-Dimethylcyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	24	LC50	6200	Benville et al., 1985
1,3-Dimethylcyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	96	LC50	6200	Benville et al., 1985
1,3-Dimethylcyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	24	LC50	6200	Benville et al., 1985
1,3-Dimethylcyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	96	LC50	6200	Benville et al., 1985
1,3-Dimethylcyclohexane	<i>Morone saxatilis</i>	Striped bass	24	LC50	9900	Benville et al., 1985
1,3-Dimethylcyclohexane	<i>Morone saxatilis</i>	Striped bass	96	LC50	9300	Benville et al., 1985
1,3-Dimethylcyclohexane	<i>Morone saxatilis</i>	Striped bass	24	LC50	9900	Benville et al., 1985
1,3-Dimethylcyclohexane	<i>Morone saxatilis</i>	Striped bass	96	LC50	9300	Benville et al., 1985
1,4-Dimethylcyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	24	LC50	2800	Benville et al., 1985
1,4-Dimethylcyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	96	LC50	1500	Benville et al., 1985
1,4-Dimethylcyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	24	LC50	2800	Benville et al., 1985
1,4-Dimethylcyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	96	LC50	1500	Benville et al., 1985
Cyclohexane	<i>Artemia salina</i>	Brine shrimp	24	LC50	7322	Abernathy et al., 1986
Cyclohexane	<i>Carassius auratus</i>	Goldfish	24	LC50	42330	Pickering and Henderson, 1966
Cyclohexane	<i>Carassius auratus</i>	Goldfish	48	LC50	42330	Pickering and Henderson, 1966
Cyclohexane	<i>Carassius auratus</i>	Goldfish	96	LC50	42330	Pickering and Henderson, 1966
Cyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	24	LC50	3400	Benville et al., 1985
Cyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	96	LC50	2400	Benville et al., 1985
Cyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	96	LC50	2400	Benville and Korn, 1977

Chemical Name	Species Scientific Name	Species Common Name	Test Duration (hours)	Endpoint	Concentration (µg/L)	Reference
Cyclohexane	<i>Lepomis macrochirus</i>	Bluegill	24	LC50	42330	Pickering and Henderson, 1966
Cyclohexane	<i>Lepomis macrochirus</i>	Bluegill	48	LC50	40600	Pickering and Henderson, 1966
Cyclohexane	<i>Lepomis macrochirus</i>	Bluegill	96	LC50	34720	Pickering and Henderson, 1966
Cyclohexane	<i>Morone saxatilis</i>	Striped bass	24	LC50	8300	Benville et al., 1985
Cyclohexane	<i>Morone saxatilis</i>	Striped bass	96	LC50	8300	Benville et al., 1985
Cyclohexane	<i>Morone saxatilis</i>	Striped bass	96	LC50	8300	Benville and Korn, 1977
Cyclohexane	<i>Pimephales promelas</i>	Fathead minnow	24	LC50	35080	Pickering and Henderson, 1966
Cyclohexane	<i>Pimephales promelas</i>	Fathead minnow	24	LC50	42330	Pickering and Henderson, 1966
Cyclohexane	<i>Pimephales promelas</i>	Fathead minnow	48	LC50	35080	Pickering and Henderson, 1966
Cyclohexane	<i>Pimephales promelas</i>	Fathead minnow	48	LC50	42330	Pickering and Henderson, 1966
Cyclohexane	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	4530	Geiger et al., 1990
Cyclohexane	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	32710	Pickering and Henderson, 1966
Cyclohexane	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	42330	Pickering and Henderson, 1966
Cyclopentane	<i>Artemia salina</i>	Brine shrimp	24	LC50	19638	Abernathy et al., 1986
Dimethylcyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	96	LC50	1500	Benville and Korn, 1977
Dimethylcyclohexane	<i>Morone saxatilis</i>	Striped bass	96	LC50	3200	Benville and Korn, 1977
Ethylcyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	24	LC50	3700	Benville et al., 1985
Ethylcyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	96	LC50	3100	Benville et al., 1985
Hexane	<i>Artemia salina</i>	Brine shrimp	24	LC50	3533	Abernathy et al., 1986
Hexane	<i>Pimephales promelas</i>	Fathead minnow	96	LC50	2500	Geiger et al., 1990
Methylcyclohexane	<i>Artemia salina</i>	Brine shrimp	24	LC50	3682	Abernathy et al., 1986
Methylcyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	24	LC50	3500	Benville et al., 1985
Methylcyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	96	LC50	3300	Benville et al., 1985
Methylcyclohexane	<i>Crangon franciscorum</i>	Bay shrimp	96	LC50	3500	Benville and Korn, 1977
Methylcyclohexane	<i>Morone saxatilis</i>	Striped bass	24	LC50	7000	Benville et al., 1985
Methylcyclohexane	<i>Morone saxatilis</i>	Striped bass	96	LC50	5800	Benville et al., 1985
Methylcyclohexane	<i>Morone saxatilis</i>	Striped bass	96	LC50	5800	Benville and Korn, 1977
Octane	<i>Artemia salina</i>	Brine shrimp	24	LC50	400	Abernathy et al., 1986
Pentane	<i>Artemia salina</i>	Brine shrimp	24	LC50	11905	Abernathy et al., 1986

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Appendix B
Calculation Spreadsheets

Table B-1. Spreadsheet for Derivation of Risk-based Sediment Benchmarks for Petroleum Hydrocarbons

Fraction	Hydrocarbons	Solubility, 25°C (mg/L)	Log K _{ow}	Geomean Log K _{ow}	Log LC50 ^a (mM/L)	LC50 (mM/L)	Molecular Weight	LC50 (mg/L)	>Solubility (y/n)	Geomean LC50 (mg/L)	FCV ^b (mg/L)	FCV (ug/L)	Log K _{oc} ^c	Sediment Benchmark (ug/g oc) ^d	Sediment Benchmark (f=0.001) (ug/g) ^d
C ₅ -C ₈	Aliphatics														
	n-Pentane	38	3.45	4.12	-0.85	1.42E-01	72.15	10.24	n	3.2654	0.2177	218	3.86	1591	1.59
	2,2-Dimethylbutane	18.4	3.82		-1.18	6.62E-02	86.18	5.71	n						
	Cyclopentane	156	3		-0.44	3.59E-01	70.14	25.18	n						
	2,3-Dimethylbutane	19.1	3.85		-1.21	6.22E-02	86.18	5.36	n						
	2-Methylpentane	13.8	3.74		-1.11	7.81E-02	86.18	6.73	n						
	3-Methylpentane	12.8	3.6		-0.98	1.04E-01	86.18	8.98	n						
	n-Hexane	9.5	4.11		-1.44	3.64E-02	86.18	3.14	n						
	cycloheptane	30	4		-1.34	4.57E-02	98.19	4.49	n						
	2,2-Dimethylpentane	4.4	4.14		-1.47	3.42E-02	100.2	3.43	n						
	Methylcyclopentane	42	3.37		-0.78	1.67E-01	84.16	14.09	n						
	2,4-Dimethylpentane	4.06	4.14		-1.47	3.42E-02	100.2	3.43	n						
	2,2,3-Trimethylbutane	4.38	4.03		-1.37	4.29E-02	100.2	4.30	n						
	3,3-Dimethylpentane	5.94	4.14		-1.47	3.42E-02	100.2	3.43	n						
	Cyclohexane	55	3.44		-0.84	1.45E-01	84.16	12.20	n						
	2,3-Dimethylpentane	5.25	4.14		-1.47	3.42E-02	100.2	3.43	n						
	3-Methylhexane	3.3	4.27		-1.58	2.62E-02	100.2	2.62	n						
	2,2,4-Trimethylpentane	2.44	4.54		-1.82	1.50E-02	114.2	1.71	n						
	Heptane	2.9	5		-2.24	5.81E-03	100.2	0.58	n						
	Methylcyclohexane	14	3.88		-1.23	5.85E-02	98.19	5.75	n						
	1,1,3-Trimethylcyclopentane	3.73	4.35		-1.65	2.22E-02	112.2	2.49	n						
	2,3,4-Trimethylpentane	2	4.54		-1.82	1.50E-02	114.2	1.71	n						
	2-Methylheptane	0.85	4.8		-2.06	8.78E-03	114.2	1.00	y						
	3-Methylheptane	0.79	4.8		-2.06	8.78E-03	114.2	1.00	y						
	1,4-Dimethylcyclohexane	3.84	4.39		-1.69	2.04E-02	112.2	2.29	n						
	2,2,5-Trimethylhexane	1.15	5.06		-2.29	5.14E-03	128.3	0.66	n						
	1,2-Dimethylcyclohexane	6	4.39		-1.69	2.04E-02	112.2	2.29	n						
	Octane	0.66	5.15		-2.37	4.27E-03	114.2	0.49	n						
C ₉ -C ₁₂	3-Methyloctane	1.42	5.32	6.01	-2.52	3.01E-03	128.3	0.39	n	0.09425	0.006	6.3	5.64	2722	2.72
	Nonane	0.22	5.65		-2.82	1.52E-03	128.3	0.20	n						
	Decane	0.052	6.25		-3.35	4.42E-04	142.3	0.06	y						
	Undecane	0.04	6.94		-3.97	1.07E-04	156.3	0.02	n						
C ₁₃ -C ₁₈	Tetradecane	7.00E-04	7.2	8.75	-4.21	6.24E-05	198.4	0.0124	y	0.0007561	5.04E-05	0.05	8.04	5543	5.54
	Pentadecane	8.00E-05	8.63		-5.49	3.27E-06	212.4	0.0007	y						
	Hexadecane	5.00E-05	8.25		-5.15	7.16E-06	226.4	0.0016	y						
	Heptadecane	6.00E-06	9.69		-6.43	3.68E-07	240.5	0.0001	y						
	Octadecane	4.00E-06	9.32		-6.10	7.89E-07	254.5	0.0002	y						

Fraction	Hydrocarbons	Solubility, 25°C (mg/L)	Log K _{ow}	Geomean Log K _{ow}	Log LC50 ^a (mM/L)	LC50 (mM/L)	Molecular Weight	LC50 (mg/L)	>Solubility (y/n)	Geomean LC50 (mg/L)	FCV ^b (mg/L)	FCV (ug/L)	Log K _{oc} ^c	Sediment Benchmark (ug/g oc) ^d	Sediment Benchmark (f=0.001) (ug/g) ^d
C ₁₉ -C ₃₆	Nonadecane	4.00E-07	10.72	11.64	-7.36	4.40E-08	268.5	1.18E-05	y	1.77E-06	1.18E-07	0.0001	10.92	9883	9.88
	Eicosane	3.00E-07	11.27		-7.85	1.42E-08	282.6	4.00E-06	y						
	Tetracosane	6.00E-10	13.07		-9.46	3.46E-10	338.7	1.17E-07	y						
C ₆ -C ₈	Benzene	1770	2.13	2.82	0.31	2.03	78.1	158.18	n	17.863	1.19	1191	2.65	531	0.53
	Toluene	530	2.69		-0.50	0.318	92.1	29.33	n						
	Ethylbenzene	169	3.13		-1.13	7.44E-02	106.2	7.91	n						
	p-Xylene	180	3.18		-1.20	6.31E-02	106.2	6.7	n						
	o-Xylene	220	3.15		-1.16	6.97E-02	106.2	7.4	n						
C ₉ -C ₁₂	Isopropylbenzene	56	3.63	3.94	-1.85	1.43E-02	120.2	1.72	n	0.6923	0.0462	46.2	3.69	228	0.23
	n-Propylbenzene	55	3.69		-1.93	1.17E-02	120.2	1.41	n						
	1-Methyl-4-ethylbenzene	95	3.63		-1.85	1.43E-02	120.2	1.72	n						
	1,3,5-Trimethylbenzene	48.9	3.58		-1.77	1.68E-02	120.2	2.02	n						
	Naphthalene	31	3.37		-1.47	3.37E-02	128.2	4.32	n						
	2-Methylnaphthalene	25	3.86		-2.18	6.68E-03	142.2	0.95	n						
	1-Methylnaphthalene	28	3.87		-2.19	6.46E-03	142.2	0.92	n						
	1-Methyl-4-isopropylbenzene	34	4.1		-2.52	3.02E-03	134.2	0.41	n						
	2-Ethyl-naphthalene	8	4.4		-2.95	1.12E-03	156.2	0.18	n						
	Biphenyl	7.2	3.69		-1.93	1.17E-02	154.2	1.80	n						
	Isobutylbenzene	10.1	4.01		-2.39	4.07E-03	134.2	0.55	n						
	n-Butylbenzene	15	4.26		-2.75	1.78E-03	134.2	0.24	n						
	Sec-Butylbenzene	14	4.1		-2.52	3.02E-03	134.2	0.41	n						
	1,2,4,5-Tetramethylbenzene	3.48	4.1		-2.52	3.02E-03	134.2	0.41	n						
	2,6-Dimethylnaphthalene	1.7	4.1		-2.52	3.02E-03	156.2	0.47	n						
	2,3-Dimethylnaphthalene	2.5	4.4		-2.95	1.12E-03	156.2	0.18	n						
	Dibenzothiophene	1.47	4.38		-2.92	1.20E-03	184.3	0.22	n						
	Acenaphthene	3.8	3.92		-2.26	5.47E-03	154.2	0.84	n						
C ₁₃ -C ₁₅	Fluorene	1.9	3.97	4.67	-2.33	4.64E-03	166.2	0.77	n	0.07758	0.0052	5.2	4.38	125	0.13
	Phenanthrene	1.1	4.36		-2.89	1.28E-03	178.2	0.23	n						
	2-Methylanthracene	0.03	5.15		-4.03	9.41E-05	192.3	0.02	n						
	9-Methylanthracene	0.261	5.07		-3.91	1.23E-04	192.3	0.02	n						
	1-Methylphenanthrene	0.27	4.93		-3.71	1.95E-04	192.3	0.04	n						
C ₁₆ -C ₂₄	Benzo(a)fluorene	0.045	5.4	5.9	-4.38	4.12E-05	216.3	0.0089	n	0.001746	0.00012	0.12	5.53	40	0.04
	Benz(a)anthracene	0.011	5.91		-5.12	7.64E-06	228.3	0.0017	n						
	Chrysene	1.90E-03	5.79		-4.94	1.14E-05	228.3	0.0026	y						
	Benzo(a)pyrene	1.00E-03	6.04		-5.30	4.98E-06	252.3	0.0013	y						
	Pyrene	0.13	5.18		-4.07	8.52E-05	202.3	0.02	n						
	Fluoranthene	0.24	5.22		-4.13	7.47E-05	202.3	0.02	n						

Fraction	Hydrocarbons	Solubility, 25°C (mg/L)	Log K _{ow}	Geomean Log K _{ow}	Log LC50 ^a (mM/L)	LC50 (mM/L)	Molecular Weight	LC50 (mg/L)	>Solubility (y/n)	Geomean LC50 (mg/L)	FCV ^b (mg/L)	FCV (ug/L)	Log K _{oc} ^c	Sediment Benchmark (ug/g oc) ^d	Sediment Benchmark (f=0.001) (ug/g) ^d
	Coronene	1.4E-04	6.75		-6.32	4.77E-07	300.4	0.0001	y						
	Benzo(ghi)perylene	1.40E-04	6.29		-5.66	2.18E-06	276.3	0.0006	y						
	Dibenz(a,h)anthracene	5.80E-04	6.75		-6.32	4.77E-07	278.4	0.0001	n						

^a Values obtained from regression equation and corresponding log K_{ow}

^b To obtain the final chronic aquatic value, the mean LC50 was divided by an ACR of 15

^c Calculated from the equation: Log K_{oc} = 0.00028 + 0.938Log K_{ow}

^d µg/g is equivalent to mg/kg or parts per million (ppm)

Table B-2. Spreadsheet for Derivation of Sediment Benchmarks for Petroleum Hydrocarbons – Current MADEP Fractions

Fraction	Hydrocarbons	Solubility, 25°C (mg/L)	Log K _{ow}	Geomean Log K _{ow}	Log LC50 ^a (mM/L)	LC50 (mM/L)	Molecular Weight	LC50 (mg/L)	>Solubility (y/n)	Geomean LC50 (mg/L)	FCV ^b (mg/L)	FCV (ug/L)	Log K _{oc} ^c	Sediment Benchmark (ug/g oc) ^d	Sediment Benchmark (f=0.001) (ug/g) ^d
C ₅ -C ₈	Aliphatics														
	n-Pentane	38	3.45	4.12	-0.85	1.42E-01	72.15	10.24	n	3.2654	0.2177	218	3.86	1591	1.59
	2,2-Dimethylbutane	18.4	3.82		-1.18	6.62E-02	86.18	5.71	n						
	Cyclopentane	156	3		-0.44	3.59E-01	70.14	25.18	n						
	2,3-Dimethylbutane	19.1	3.85		-1.21	6.22E-02	86.18	5.36	n						
	2-Methylpentane	13.8	3.74		-1.11	7.81E-02	86.18	6.73	n						
	3-Methylpentane	12.8	3.6		-0.98	1.04E-01	86.18	8.98	n						
	n-Hexane	9.5	4.11		-1.44	3.64E-02	86.18	3.14	n						
	cycloheptane	30	4		-1.34	4.57E-02	98.19	4.49	n						
	2,2-Dimethylpentane	4.4	4.14		-1.47	3.42E-02	100.2	3.43	n						
	Methylcyclopentane	42	3.37		-0.78	1.67E-01	84.16	14.09	n						
	2,4-Dimethylpentane	4.06	4.14		-1.47	3.42E-02	100.2	3.43	n						
	2,2,3-Trimethylbutane	4.38	4.03		-1.37	4.29E-02	100.2	4.30	n						
	3,3-Dimethylpentane	5.94	4.14		-1.47	3.42E-02	100.2	3.43	n						
	Cyclohexane	55	3.44		-0.84	1.45E-01	84.16	12.20	n						
	2,3-Dimethylpentane	5.25	4.14		-1.47	3.42E-02	100.2	3.43	n						
	3-Methylhexane	3.3	4.27		-1.58	2.62E-02	100.2	2.62	n						
	2,2,4-Trimethylpentane	2.44	4.54		-1.82	1.50E-02	114.2	1.71	n						
	Heptane	2.9	5		-2.24	5.81E-03	100.2	0.58	n						
	Methylcyclohexane	14	3.88		-1.23	5.85E-02	98.19	5.75	n						
	1,1,3-Trimethylcyclopentane	3.73	4.35		-1.65	2.22E-02	112.2	2.49	n						
	2,3,4-Trimethylpentane	2	4.54		-1.82	1.50E-02	114.2	1.71	n						
	2-Methylheptane	0.85	4.8		-2.06	8.78E-03	114.2	1.00	y						
	3-Methylheptane	0.79	4.8		-2.06	8.78E-03	114.2	1.00	y						
	1,4-Dimethylcyclohexane	3.84	4.39		-1.69	2.04E-02	112.2	2.29	n						
	2,2,5-Trimethylhexane	1.15	5.06		-2.29	5.14E-03	128.3	0.66	n						
	1,2-Dimethylcyclohexane	6	4.39		-1.69	2.04E-02	112.2	2.29	n						
	Octane	0.66	5.15		-2.37	4.27E-03	114.2	0.49	n						
C ₉ -C ₁₈	3-Methyloctane	1.42	5.32	7.32	-2.52	3.01E-03	128.3	0.39	n	0.00646	0.0004	0.4	6.87	3167	3.17
	Nonane	0.22	5.65		-2.82	1.52E-03	128.3	0.20	n						
	Decane	0.052	6.25		-3.35	4.42E-04	142.3	0.06	y						
	Undecane	0.04	6.94		-3.97	1.07E-04	156.3	0.02	n						
	Tetradecane	7.00E-04	7.2		-4.21	6.24E-05	198.4	0.0124	y						
	Pentadecane	8.00E-05	8.63		-5.49	3.27E-06	212.4	0.0007	y						
	Hexadecane	5.00E-05	8.25		-5.15	7.16E-06	226.4	0.0016	y						
	Heptadecane	6.00E-06	9.69		-6.43	3.68E-07	240.5	0.0001	y						
	Octadecane	4.00E-06	9.32		-6.10	7.89E-07	254.5	0.0002	y						

Fraction	Hydrocarbons	Solubility, 25°C (mg/L)	Log K _{ow}	Geomean Log K _{ow}	Log LC50 ^a (mM/L)	LC50 (mM/L)	Molecular Weight	LC50 (mg/L)	>Solubility (y/n)	Geomean LC50 (mg/L)	FCV ^b (mg/L)	FCV (ug/L)	Log K _{oc} ^c	Sediment Benchmark (ug/g oc) ^d	Sediment Benchmark (f=0.001) (ug/g) ^d
C ₁₉ -C ₃₆	Nonadecane	4.00E-07	10.72	11.64	-7.36	4.40E-08	268.5	1.18E-05	y	1.77E-06	1.18E-07	0.0001	10.92	9883	9.88
	Eicosane	3.00E-07	11.27		-7.85	1.42E-08	282.6	4.00E-06	y						
	Tetracosane	6.00E-10	13.07		-9.46	3.46E-10	338.7	1.17E-07	y						
C ₉ -C ₁₀	Isopropylbenzene	56	3.63	3.84	-1.85	1.43E-02	120.2	1.72	n	0.8907	0.0594	59.4	3.60	236	0.24
	n-Propylbenzene	55	3.69		-1.93	1.17E-02	120.2	1.41	n						
	1-Methyl-4-ethylbenzene	95	3.63		-1.85	1.43E-02	120.2	1.72	n						
	1,3,5-Trimethylbenzene	48.9	3.58		-1.77	1.68E-02	120.2	2.02	n						
	Naphthalene	31	3.37		-1.47	3.37E-02	128.2	4.32	n						
	1-Methyl-4-isopropylbenzene	34	4.1		-2.52	3.02E-03	134.2	0.41	n						
	Isobutylbenzene	10.1	4.01		-2.39	4.07E-03	134.2	0.55	n						
	n-Butylbenzene	15	4.26		-2.75	1.78E-03	134.2	0.24	n						
	Sec-Butylbenzene	14	4.1		-2.52	3.02E-03	134.2	0.41	n						
	1,2,4,5-Tetramethylbenzene	3.48	4.1		-2.52	3.02E-03	134.2	0.41	n						
C ₁₁ -C ₂₂	2-Methylnaphthalene	25	3.86	4.81	-2.18	6.68E-03	142.2	0.95	n	0.0421	0.0028	2.8	4.52	92	0.09
	1-Methylnaphthalene	28	3.87		-2.19	6.46E-03	142.2	0.92	n						
	2-Ethylnaphthalene	8	4.4		-2.95	1.12E-03	156.2	0.18	n						
	Biphenyl	7.2	3.69		-1.93	1.17E-02	154.2	1.80	n						
	2,6-Dimethylnaphthalene	1.7	4.1		-2.52	3.02E-03	156.2	0.47	n						
	2,3-Dimethylnaphthalene	2.5	4.4		-2.95	1.12E-03	156.2	0.18	n						
	Dibenzothiophene	1.47	4.38		-2.92	1.20E-03	184.3	0.22	n						
	Acenaphthene	3.8	3.92		-2.26	5.47E-03	154.2	0.84	n						
	Fluorene	1.9	3.97		-2.33	4.64E-03	166.2	0.77	n						
	Phenanthrene	1.1	4.36		-2.89	1.28E-03	178.2	0.23	n						
	2-Methylanthracene	0.03	5.15		-4.03	9.41E-05	192.3	0.02	n						
	9-Methylanthracene	0.261	5.07		-3.91	1.23E-04	192.3	0.02	n						
	1-Methylphenanthrene	0.27	4.93		-3.71	1.95E-04	192.3	0.04	n						
	Benzo(a)fluorene	0.045	5.4		-4.38	4.12E-05	216.3	0.0089	n						
	Benz(a)anthracene	0.011	5.91		-5.12	7.64E-06	228.3	0.0017	n						
	Chrysene	1.90E-03	5.79		-4.94	1.14E-05	228.3	0.0026	y						
	Benzo(a)pyrene	1.00E-03	6.04		-5.30	4.98E-06	252.3	0.0013	y						
	Pyrene	0.13	5.18		-4.07	8.52E-05	202.3	0.02	n						
	Fluoranthene	0.24	5.22		-4.13	7.47E-05	202.3	0.02	n						
	Benzo(ghi)perylene	1.40E-04	6.29		-5.66	2.18E-06	276.3	0.0006	y						
	Dibenz(a,h)anthracene	5.80E-04	6.75		-6.32	4.77E-07	278.4	0.0001	n						

^a Values obtained from regression equation and corresponding log K_{ow}

^b To obtain the final chronic aquatic value, the mean LC50 was divided by an ACR of 15

^c Calculated from the equation: Log K_{oc} = 0.00028 + 0.938Log K_{ow}

^d µg/g is equivalent to mg/kg or parts per million (ppm)